

# ICSE Class 10 Chemistry Question Paper Solution 2017

## CHEMISTRY (SCIENCE PAPER-2)

### SECTION I (40 Marks)

*Attempt all questions from this Section*

#### Question 1

- (a) Fill in the blanks from the choices given in brackets: [5]
- (i) The energy required to remove an electron from a neutral isolated gaseous atom and convert it into a positively charged gaseous ion is called \_\_\_\_\_.  
(*electron affinity, ionisation potential, electronegativity*)
  - (ii) The compound that does not have a lone pair of electrons is \_\_\_\_\_.  
(*water, ammonia, carbon tetra chloride*)
  - (iii) When a metallic oxide is dissolved in water, the solution formed has a high concentration of \_\_\_\_\_ ions. ( *$H^+$ ,  $H_3O^+$ ,  $OH^-$* )
  - (iv) Potassium sulphite on reacting with hydrochloric acid releases \_\_\_\_\_.  
( *$Cl_2$ ,  $SO_2$ ,  $H_2S$* )
  - (v) The compound formed when ethene reacts with Hydrogen is \_\_\_\_\_.  
( *$CH_4$ ,  $C_2H_6$ ,  $C_3H_8$* )
- (b) Choose the **correct answer** from the options given below: [5]
- (i) A **chloride** which forms a precipitate that is soluble in excess of ammonium hydroxide, is:
    - 1. Calcium chloride
    - 2. Ferrous chloride
    - 3. Ferric chloride
    - 4. Copper chloride
  - (ii) If the molecular formula of an organic compound is  $C_{10}H_{18}$  it is:
    - 1. alkene

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**LL Leslie**



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