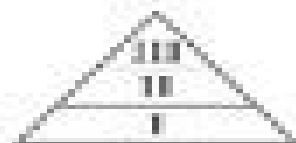


Name:

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1. An energy pyramid containing green plants and other organisms from a food chain is represented below. Herbivores would most likely be located in



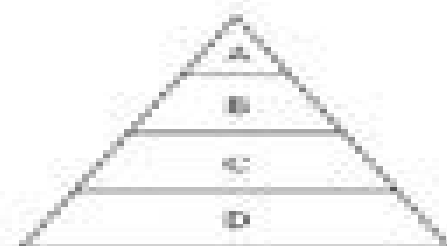
(A) level I, only

(B) level II, only

(C) level III, only

(D) level I and level II

2. The diagram below represents an energy pyramid.



At which level do the organisms all carry on both respiration and photosynthesis?

(A) A (B) B

(C) C (D) D

3. As energy moves through a forest ecosystem, it flows from

(A) heterotrophs to autotrophs

(B) animals to plants

(C) herbivores to carnivores

(D) carnivores to autotrophs

4. The diagram represents the energy relationships in a forest ecosystem.

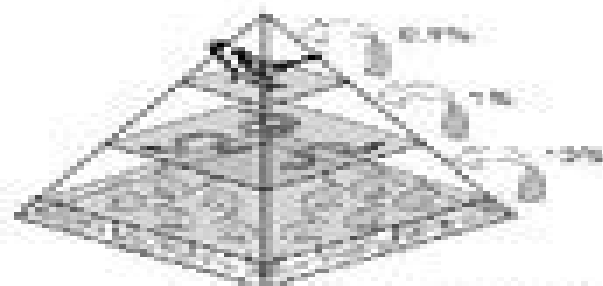


Figure Adapted from *OpenStax*, *Biology*, 2019

The best explanation for the decrease in the amount of energy available as one moves up the pyramid is that

(A) producers require more energy than consumers to survive

(B) decomposers recycle nutrients at each level

(C) much of the energy at each level is lost as heat

(D) animals use less energy than plants

5. Evidence suggests that a large meteorite hit Earth 65 million years ago, causing a layer of dust to block the Sun, cooling the planet. It is estimated that 70% of all plant and animal species, including the dinosaurs, died off as a result. The best explanation for the deaths of these organisms is that

(A) consumers require sunlight to make sugars in order to survive

(B) the amount of energy available to the biosphere was decreased

(C) energy is produced only by plants

(D) all animals eat plants for energy

Energy Pyramids Pearson Education Answers

Katrin Zwirgmaier



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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at a given pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 °C and 64 °C respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the powdered acetamide occupies 1-2 mm of the capillary tube end c

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affected by three key factors the first being intermolecular forces intermolecular forces affect boiling point based on bond strength

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is

at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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