

## Empirical and Molecular Formula Worksheet ANSWER KEY

Write the empirical formula for the following compounds.

- 1)  $C_6H_6$     **CH**
- 6)  $C_8H_{14}$      **$C_4H_7$**
- 7)  $WO_2$      **$WO_2$**
- 8)  $C_2H_6O_2$      **$CH_3O$**
- 9)  $X_{10}Y_{12}$      **$X_5Y_6$**
- 6) A compound with an empirical formula of  $C_2OH_4$  and a molar mass of 88 grams per mole. What is the molecular formula of this compound?     **$C_8O_4H_{16}$**
- 7) A compound with an empirical formula of  $C_4H_4O$  and a molar mass of 136 grams per mole. What is the molecular formula of this compound?     **$C_8H_8O_2$**
- 8) A compound with an empirical formula of  $CFBrO$  and a molar mass of 253.8 grams per mole. What is the molecular formula of this compound?     **$C_2F_2Br_2O_2$**
- 9) A compound with an empirical formula of  $C_3H_3N$  and a molar mass of 46 grams per mole. What is the molecular formula of this compound?     **$C_6H_6N_2$**
- 10) A well-known reagent in analytical chemistry, dimethylglyoxime, has the empirical formula  $C_2H_5NO$ . If its molar mass is 116.1 g/mol, what is the molecular formula of the compound?     **$C_8H_{10}N_2O_2$**
12. A certain blue solid contains 36.84% N and 63.16% O. What is the empirical formula of this compound? **The ratios are  $N_{1.50}O_{1.50}$ . Since 1.50 is not close to a whole number, we multiply *both* subscripts by 2. The empirical formula is thus  $N_2O_2$ . (The name is dinitrogen trioxide.)**
13. A sample of indium chloride weighing 0.5000 g is found to contain 0.2404 g of chlorine. What is the empirical formula of the indium compound?     **$InCl_2$**
14. An unknown compound was found to have a percent composition as follows: 47.0 % potassium, 14.5 % carbon, and 38.5 % oxygen. What is its empirical formula? If the true molar mass of the compound is 166.22 g/mol, what is its molecular formula?     **$K_2C_3O_4$**
15. Rubbing alcohol was found to contain 60.0 % carbon, 13.4 % hydrogen, and the remaining mass was due to oxygen. What is the empirical formula of rubbing alcohol?     **$C_3H_8O$**

# Empirical And Molecular Formula Quiz With Answers

**Annelies Wilder-Smith**



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## **Table of Contents Empirical And Molecular Formula Quiz With Answers**

1. Understanding the eBook Empirical And Molecular Formula Quiz With Answers
  - The Rise of Digital Reading Empirical And Molecular Formula Quiz With Answers
  - Advantages of eBooks Over Traditional Books
2. Identifying Empirical And Molecular Formula Quiz With Answers
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Empirical And Molecular Formula Quiz With Answers
  - User-Friendly Interface
4. Exploring eBook Recommendations from Empirical And Molecular Formula Quiz With Answers
  - Personalized Recommendations
  - Empirical And Molecular Formula Quiz With Answers User Reviews and Ratings
  - Empirical And Molecular Formula Quiz With Answers and Bestseller Lists
5. Accessing Empirical And Molecular Formula Quiz With Answers Free and Paid eBooks
  - Empirical And Molecular Formula Quiz With Answers Public Domain eBooks
  - Empirical And Molecular Formula Quiz With Answers eBook Subscription Services
  - Empirical And Molecular Formula Quiz With Answers Budget-Friendly Options

6. Navigating Empirical And Molecular Formula Quiz With Answers eBook Formats
  - ePub, PDF, MOBI, and More
  - Empirical And Molecular Formula Quiz With Answers Compatibility with Devices
  - Empirical And Molecular Formula Quiz With Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Empirical And Molecular Formula Quiz With Answers
  - Highlighting and Note-Taking Empirical And Molecular Formula Quiz With Answers
  - Interactive Elements Empirical And Molecular Formula Quiz With Answers
8. Staying Engaged with Empirical And Molecular Formula Quiz With Answers
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Empirical And Molecular Formula Quiz With Answers
9. Balancing eBooks and Physical Books Empirical And Molecular Formula Quiz With Answers
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Empirical And Molecular Formula Quiz With Answers
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Empirical And Molecular Formula Quiz With Answers
  - Setting Reading Goals Empirical And Molecular Formula Quiz With Answers
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Empirical And Molecular Formula Quiz With Answers
  - Fact-Checking eBook Content of Empirical And Molecular Formula Quiz With Answers
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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