

BOY'S GAS CALORIMETER

AIM:- To determine the calorific value of the gaseous fuel.

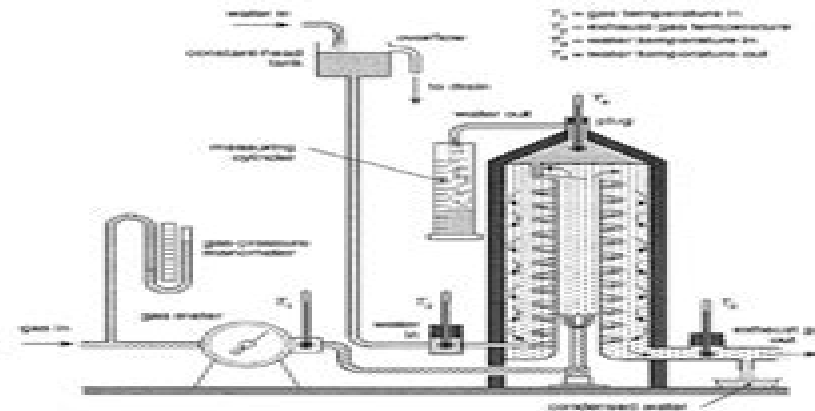
THEORY:- calorific value of solid fuel is the amount heat released when 1kg of fuel is burnt, completely expressed in K.J/kg (complete combustion of fuel)

Calorific value of gaseous fuel is the amount heat released by 1 m³ of gaseous fuel at NTP when it is burnt (complete combustion of fuel) expressed K.J/m³at NTP

APPARATUS:- It consists of a central funnel in the middle of which, and at the bottom, is filled with burners. The funnel is fitted with cooling coil around it. The cooling water inlet passes through the calorimeter at the bottom and the outlet is at the top. Thermometers are installed to measure the inlet and outlet temperature of water. The gas is supplied at the bottom and the outlet is at the top. Water is circulated through the coil at constant head. Gas governor is provided to supply the gas at constant pressure

PROCEDURE:-

- Check the flow rate of water and gas.
- Allow the gas through the burner for initial value by adjusting the control valve.
- Put on the burner and raise the burner into the calorimeter. Allow the gas to burn until steady state is reached.
- Note down the temperature of the water at inlet and outlet



Energy Conversion Lab Manual

ML Yell



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Table of Contents Energy Converstion Lab Manual

1. Understanding the eBook Energy Converstion Lab Manual
 - The Rise of Digital Reading Energy Converstion Lab Manual
 - Advantages of eBooks Over Traditional Books
2. Identifying Energy Converstion Lab Manual
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Energy Converstion Lab Manual
 - User-Friendly Interface
4. Exploring eBook Recommendations from Energy Converstion Lab Manual
 - Personalized Recommendations
 - Energy Converstion Lab Manual User Reviews and Ratings
 - Energy Converstion Lab Manual and Bestseller Lists
5. Accessing Energy Converstion Lab Manual Free and Paid eBooks
 - Energy Converstion Lab Manual Public Domain eBooks
 - Energy Converstion Lab Manual eBook Subscription Services
 - Energy Converstion Lab Manual Budget-Friendly Options

6. Navigating Energy Conversion Lab Manual eBook Formats
 - ePub, PDF, MOBI, and More
 - Energy Conversion Lab Manual Compatibility with Devices
 - Energy Conversion Lab Manual Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Energy Conversion Lab Manual
 - Highlighting and Note-Taking Energy Conversion Lab Manual
 - Interactive Elements Energy Conversion Lab Manual
8. Staying Engaged with Energy Conversion Lab Manual
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Energy Conversion Lab Manual
9. Balancing eBooks and Physical Books Energy Conversion Lab Manual
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Energy Conversion Lab Manual
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Energy Conversion Lab Manual
 - Setting Reading Goals Energy Conversion Lab Manual
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Energy Conversion Lab Manual
 - Fact-Checking eBook Content of Energy Conversion Lab Manual
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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mention date has been approved your re joining date should be mention date