

### Assessment

## Chapter Test B

## Atomic Physics

## MULTIPLE CHOICE

In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question.

1. What is the frequency of a photon with an energy of  $1.99 \times 10^{-19} \text{ J}$ ?  
( $h = 6.63 \times 10^{-34} \text{ J}\cdot\text{s}$ )  
a.  $1.00 \times 10^{14} \text{ Hz}$   
b.  $2.00 \times 10^{14} \text{ Hz}$   
c.  $3.00 \times 10^{14} \text{ Hz}$   
d.  $4.00 \times 10^{14} \text{ Hz}$
2. Light with an energy equal to three times the work function of a given metal causes the metal to eject photoelectrons. What is the ratio of the maximum photoelectron kinetic energy to the work function?  
a. 1 : 1  
b. 2 : 1  
c. 3 : 1  
d. 4 : 1
3. A monochromatic light beam with a quantum energy value of 3.0 eV is incident upon a photocell. The work function of the photocell is 1.6 eV. What is the maximum kinetic energy of the ejected electrons?  
a. 4.6 eV  
b. 4.8 eV  
c. 1.4 eV  
d. 2.4 eV
4. What causes the bright lines in the emission spectrum of an element to occur?  
a. Photons are absorbed when electrons jump from a higher-energy to a lower-energy state.  
b. Photons are emitted when electrons jump from a higher-energy to a lower-energy state.  
c. Photons are absorbed when electrons jump from a lower-energy to a higher-energy state.  
d. Photons are emitted when electrons jump from a lower-energy to a higher-energy state.
5. What causes the dark lines in the absorption spectrum of an element to occur?  
a. Photons are absorbed when electrons jump from a higher-energy to a lower-energy state.  
b. Photons are emitted when electrons jump from a higher-energy to a lower-energy state.  
c. Photons are absorbed when electrons jump from a lower-energy to a higher-energy state.  
d. Photons are emitted when electrons jump from a lower-energy to a higher-energy state.

# Energy Chapter Physical Science

**Scott C. Dulebohn**



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section learning objectives by the end of this section you will be able to do the following describe simple and complex machines calculate mechanical advantage and efficiency of

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two or more simple machines operate together to form a compound machine study with quizlet and memorize flashcards containing terms like simple machines lever wheel and axle

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how well do you know your simple machines complete both sides of this chart define it what is it describe it what does it do what happens to the force explain your answer

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answer key a wedge has a slanted edge like an inclined plane the difference is in how it s used an inclined plane is used to move things from here to there while a wedge breaks

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procedure answer the following questions regarding simple machine systems each question requires proper illustration and annotation including labeling of forces distances direction

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google classroom there are six simple machines they are as follows lever wheel pulley incline plane wedge and screw this tutorial will cover the first three simple machines can

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across 1 simple machine that has a straight slanted surface 4 most common type of lever 5 all inclined planes must have this kind of angle 7 the fe fr 8 made up of a twisted inclined

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