

Empirical and Molecular Formula Practice Solutions

1. A compound is found to contain 63.52 % iron and 36.48% sulfur. Find its empirical formula.

$$63.52 \text{ g Fe} \times \frac{1 \text{ mol}}{55.85 \text{ g}} = 1.137 \text{ mol Fe}$$

$$36.48 \text{ g S} \times \frac{1 \text{ mol}}{32.06 \text{ g}} = 1.138 \text{ mol S}$$

$$\frac{1.137 \text{ mol Fe}}{1.137} : \frac{1.138 \text{ mol S}}{1.137} = 1 \text{ mol Fe: } 1.001 \text{ mol S}$$

empirical formula = FeS

2. A compound is found to contain 26.56% potassium, 35.41% chromium, and the remainder oxygen. Find its empirical formula.

$$26.56 \text{ g K} \times \frac{1 \text{ mol}}{39.10 \text{ g}} = 0.6793 \text{ mol K}$$

$$36.48 \text{ g Cr} \times \frac{1 \text{ mol}}{52.00 \text{ g}} = 0.7015 \text{ mol Cr}$$

$$36.96 \text{ g O} \times \frac{1 \text{ mol}}{16.00 \text{ g}} = 2.310 \text{ mol O}$$

$$\frac{0.6793 \text{ mol K}}{0.6793} : \frac{0.7015 \text{ mol Cr}}{0.6793} : \frac{2.310 \text{ mol O}}{0.6793} = 1 \text{ mol K: } 1.033 \text{ mol Cr: } 3.401 \text{ mol O}$$

$$\rightarrow 2 \text{ mol K: } 2.066 \text{ mol Cr: } 6.802 \text{ mol O}$$

empirical formula = $\text{K}_2\text{Cr}_2\text{O}_7$

3. A 60.00 g sample of tetraethyllead, a gasoline additive, is found to contain 38.43 g lead, 17.83 g carbon, and 3.74 g hydrogen. Find its empirical formula.

$$38.43 \text{ g Pb} \times \frac{1 \text{ mol}}{207.20 \text{ g}} = 0.1855 \text{ mol Pb}$$

$$17.83 \text{ g C} \times \frac{1 \text{ mol}}{12.01 \text{ g}} = 1.485 \text{ mol C}$$

$$3.74 \text{ g H} \times \frac{1 \text{ mol}}{1.01 \text{ g}} = 3.78 \text{ mol H}$$

$$\frac{0.1855 \text{ mol Pb}}{0.1855} : \frac{1.485 \text{ mol C}}{0.1855} : \frac{3.78 \text{ mol H}}{0.1855} = 1 \text{ mol Pb: } 8.005 \text{ mol C: } 20.4 \text{ mol H}$$

empirical formula = $\text{PbC}_8\text{H}_{20}$

Empirical And Molecular Formula Calculation Practice

Rosina Ehmann



Empirical And Molecular Formula Calculation Practice:

The Practice of Chemistry Donald J. Wink, Sharon Fetzer-Gislason, Sheila McNicholas, 2003-03 Students can't do chemistry if they can't do the math. *The Practice of Chemistry* First Edition is the only preparatory chemistry text to offer students targeted consistent mathematical support to make sure they understand how to use math especially algebra in chemical problem solving. The book's unique focus on actual chemical practice, extensive study tools, and integrated media makes *The Practice of Chemistry* the most effective way to prepare students for the standard general chemistry course and bright futures as science majors. This special PowerPoint tour of the text was created by Don Wink. http://www.bfwpub.com/pdfs/wink/POCPowerPoint_Final.ppt 832KB

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Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics. The focus is not to recap or review the theoretical concepts well described in the available texts. Instead, the topics and descriptions in this book make available specific, detailed, step-by-step methods and procedures for solving the major types of problems in general chemistry. Explanations, instructional process sequences, solved examples, and completely solved practice problems are greatly expanded, containing significantly more detail than can usually be devoted to in a comprehensive text. Many chapters also provide alternative viewpoints as an aid to understanding.

Key Features: The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester. Each is written in a specific and detailed step-by-step process for problem solving, whether mathematical or conceptual. Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts. Includes a chapter designed to eliminate confusion concerning acid-base reactions, which often persists through working with acid-base equilibrium. Many chapters provide alternative viewpoints as an aid to understanding. This book addresses a very real need for a large number of incoming freshmen in STEM fields.

Introduction to General, Organic, and Biochemistry Morris Hein, Scott Pattison, Susan Arena, Leo R. Best, 2014-01-15 The most comprehensive book available on the subject. *Introduction to General Organic and Biochemistry* 11th Edition continues its tradition of fostering the development of problem-solving skills, featuring numerous examples and coverage of current applications. Skillfully anticipating areas of difficulty and pacing the material accordingly, this readable work provides clear and logical explanations of chemical concepts as well as the right mix of general chemistry, organic chemistry, and biochemistry. An emphasis on real-world topics lets readers clearly see how the chemistry will apply to their career.

Chemistry Calculations for Beginners John Obimakinde, Samuel Obimakinde, Ebenezer Obimakinde, Fredrick Akinbolade, 2025-05-30 With decades of combined experience as science teachers at both school and undergraduate levels, the authors have recognised that one of the greatest challenges faced by

students studying chemistry is grasping the complexity of the numerous numerical problems found in most parts of the subject. This text is crafted to provide a clear and accessible pathway to overcoming this challenge by assisting students, especially novices or those with minimal knowledge of the subject, in performing chemistry calculations. The content covers fundamental calculations crucial to understanding the principles of chemistry, making it an invaluable tool for students aiming to excel in their studies.

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AP Chemistry Premium, 2025: Prep Book with 6 Practice Tests + Comprehensive Review + Online Practice Barron's Educational Series, Neil D. Jespersen, Pamela Kerrigan, 2024-07-02

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Comprehensive Chemistry XI Dr. B. Kapila, S. K. Khanna, 2010-11 Comprehensive chemistry according to the new syllabus prescribed by Central Board of Secondary Education (CBSE)

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