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Elizabeth L. Angeli

Professor Patricia Sullivan

English 624

12 February 2012

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Angeli 1

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While researching texts written about nineteenth century farming, I found a few authors who published books about the literature of nineteenth century farming, particularly agricultural journals, newspapers, pamphlets, and brochures. These authors often placed the farming literature they were studying into an historical context by

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the undergraduate reader directly Drawing on their own experiences these teachers as writers invite students to join in the larger conversation about the craft of writing Consequently each essay functions as a standalone text that can easily complement other selected readings in writing or writing intensive courses across the disciplines at any level Volume 2 continues the tradition of the previous volume with topics such as the rhetorical situation collaboration documentation styles weblogs invention writing assignment interpretation reading critically information literacy ethnography interviewing argument document design and source integration

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web exam tip you need to appreciate that the measurement of amounts in moles can apply to atoms molecules ions electrons formulae and equations e g in one mole of carbon c the number of atoms is the same as the number of molecules in one mole of carbon dioxide co₂ linking the mole and the atomic mass

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web for practical purposes a rounded version of the constant can be used in exams $6.02 \times 10^{23} \text{ mol}^{-1}$ the reason we have mol^{-1} as the unit is because this is the number of entities per mole of substance for example one mole of sodium na contains

6.02×10^{23} atoms of sodium one mole of hydrogen h_2 contains 6.02×10^{23} molecules of hydrogen