Study Guide for Content Mastery Chemistry: Matter and Change • Chapter 3

Section 3.3. Properties of Matter

Label each property as either physical or chemical

- 1. Chemical formula HJO
- 2. Forms green carbonate when exposed to moist air
- is Remains unchanged when in the presence of hitrogen-
- 4. Collord sees
- 5. Solid at normal temperatures and pressures
- 6. Ability to combine with another substance
- 7. Melting point
- Liquid at normal temperatures and pressures.
- a Boiling point is \$60°C
- 56. Conducts electricity
- 11. Density is 1.0 game

For each statement below, write true or Kelse.

- All matter that we encounter in everyday life exists in one of three physical forms.
- 13. A solid has definite shape and volume.
- 24. A liquid has a definite shape and takes on the volume of its container.
- 15. A gas has both the shape and the volume of its container.
- 16. The particles in a gas cannot be compressed into a smaller volume.
- 17. Liquide bend to contract when heated:
- 18. The particles in a solid are spaced for apart.
- 35. The words earland vapor can be used interchangeably:

Management St. 271

What kinds of changes do these words indicate? Write each word under the correct heading. Use each word only once.

1.44	And the second second	and the second	According to the second	Managedia	Committee .
Exercise 1	The second second	and the second second	-	100	
Correction to		Committee of the commit			2.00
			The second second second	3	
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		1.0			
1000 C					
46.5		1.00			
2000		3. G/86.5			
Control of the Contro		3.40			
200		2.45			
III Consultation of		100 Chillian 1997		DISSOURCE CHESTA	

For each item in Column A. write the letter of the matching item in Column B. Column A.

Columno Bi

- The new substances that are formed in a chemical reaction a chemical change
- 18. A chemical reaction that involves one or more substances. It reactions:

changing into new substances s. s. products

Glencoe Chemistry Matter And Change Study Guide Answer Key

Richard Bailey

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