

Activity A:

Subatomic particles

Get the Gizmo ready:

- Use the arrows to create an atom with two protons, two neutrons, and two electrons.
- Turn on **Show element name**.



Question: What are the properties of protons, neutrons, and electrons?

1. **Observe:** Turn on **Show element symbol** and **Element notation**. Three numbers surround the element symbol: the **mass number** (A), electrical charge (no number is displayed if the atom is neutral), and the **atomic number** (Z).
2. **Investigate:** Watch how the numbers change as you add or remove particles.



- A. Which number is equal to the number of protons in the atom?
- B. How can you calculate the number of neutrons (N) in an atom?
- C. Which particle (proton, neutron, or electron) has a
Positive charge?
Negative charge?

neutrons

by the number of protons

electron

proton

Electrical charge

3. **Observe:** Turn on **Electron configuration**. An atom's electrons form an electron cloud. Electrons are arranged in shells. The innermost shell is closest to the nucleus and contains the fewest electrons. The outermost shell is farthest from the nucleus and contains the most electrons. Electrons in the outermost shell are called valence electrons. They are responsible for the atom's chemical properties.

- A. What are the shells called?
- B. What are the shells called?
- C. What are the shells called?

Electron shells

Electron shells

Electron shells

4. **Observe:** Turn on **Electron configuration**. The following questions are asked:

- A. How many electrons are in the outermost shell of Helium?
- B. How many electrons are in the outermost shell of Hydrogen?
- C. What is the mass number of an atom with 2 protons and 2 neutrons?
- D. What is the mass number of an atom with 2 protons and 2 neutrons?
- E. How many electrons are in Helium?

2

1

4

4

2

Gizmos Element Builder Student Worksheet

Sabine Zange

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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