

### Activity A:

#### Subatomic particles

#### Get the Gizmo ready:

- Use the arrows to create an atom with two protons, two neutrons, and two electrons.
- Turn on **Show element name**.



**Question:** What are the properties of protons, neutrons, and electrons?

1. **Observe:** Turn on **Show element symbol** and **Element notation**. Three numbers surround the element symbol: the **mass number** ( $A$ ), electrical charge (no number is displayed if the atom is neutral), and the **atomic number** ( $Z$ ).
2. **Investigate:** Watch how the numbers change as you add or remove particles.



- A. Which number is equal to the number of protons in the atom?
- B. How can you calculate the number of neutrons ( $N$ ) in an atom?
- C. Which particle (proton, neutron, or electron) has a  
Positive charge?  
Negative charge?

neutrons

by the number of protons

electron

proton

neutron

3. **Explore:** Use the **Build** button to create different forms of an element. Each version of an element has the same number of protons, but a different number of neutrons. The version is represented by the atomic number and mass number. Look at the table. Some isotopes are stable, while others are **radioactive**, which means the atoms decay over time and emit radiation.

- A. What are the stable isotopes of hydrogen?
- B. What are the stable isotopes of oxygen?
- C. Carbon has three isotopes of oxygen.

1 neutron and 1 electron

7 neutrons and 7 electrons

2

4. **Explore:** Use the Gizmo to answer the following questions.

- A. How many electrons are in a neutral atom of lithium?
- B. How many electrons are in an atom of Mg-24?
- C. What is the mass number of an atom with 11 protons and 11 electrons?
- D. What is an atom's charge if it contains 11 protons and 10 electrons?
- E. How many electrons are in Mg<sup>2+</sup>?

3

12

22

1

10

# Gizmo Worksheet Answers Element Builder

**Sabine Zange**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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