

Nuclear Decay

Answer Key

Vocabulary: alpha particle, atomic number, beta particle, daughter product, gamma ray, isotope, mass number, nuclear decay, positron, radioactive, subatomic particle

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

(Note: The purpose of these questions is to activate prior knowledge and get students thinking. Students are not expected to know the answers to the Prior Knowledge Questions.)

The chart below gives the locations, charges, and approximate masses of three **subatomic particles**. The approximate mass of each particle is given in universal mass units (u).

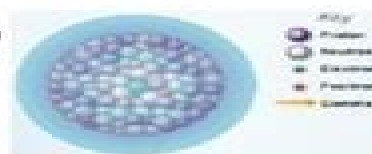
Particle	Location	Charge	Approximate mass
Proton	Nucleus	1^+	1 u
Neutron	Nucleus	0	1 u
Electron	Orbitals	1^-	0 u

- The **mass number** of an atom is equal to the sum of protons and neutrons in the nucleus. A helium atom has 2 protons and 2 neutrons. What is the mass number of this atom? 4
- The **atomic number** of an element is equal to the number of protons in each atom of the element. All helium atoms have 2 protons. What is the atomic number of helium? 2

Gizmo Warm-up

While most atoms are stable, some are **radioactive**, which means that they have a tendency to undergo spontaneous **nuclear decay**. The decay of radioactive atoms generally results in the emission of particles and/or energy.

Several types of nuclear decay can be explored with the Nuclear Decay Gizmo™. On the Gizmo, check that **Alpha decay** and **Uranium** are selected.



- Click **Play** (▶) and then click **Pause** (⏸) when the **alpha particle** is clearly visible. What is an alpha particle made of? *An alpha particle is made of 2 protons and 2 neutrons.*
- Click **Play** and observe. Besides the alpha particle, what else is emitted from the nucleus during alpha decay? *Gamma rays are also emitted.*
Gamma rays are energetic electromagnetic waves; they are often emitted in nuclear decay.

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Table of Contents Gizmo Student Exploration Nuclear Decay Answer Key

1. Understanding the eBook Gizmo Student Exploration Nuclear Decay Answer Key
 - The Rise of Digital Reading Gizmo Student Exploration Nuclear Decay Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Gizmo Student Exploration Nuclear Decay Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Gizmo Student Exploration Nuclear Decay Answer Key
 - User-Friendly Interface

4. Exploring eBook Recommendations from Gizmo Student Exploration Nuclear Decay Answer Key
 - Personalized Recommendations
 - Gizmo Student Exploration Nuclear Decay Answer Key User Reviews and Ratings
 - Gizmo Student Exploration Nuclear Decay Answer Key and Bestseller Lists
5. Accessing Gizmo Student Exploration Nuclear Decay Answer Key Free and Paid eBooks
 - Gizmo Student Exploration Nuclear Decay Answer Key Public Domain eBooks
 - Gizmo Student Exploration Nuclear Decay Answer Key eBook Subscription Services
 - Gizmo Student Exploration Nuclear Decay Answer Key Budget-Friendly Options
6. Navigating Gizmo Student Exploration Nuclear Decay Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Gizmo Student Exploration Nuclear Decay Answer Key Compatibility with Devices
 - Gizmo Student Exploration Nuclear Decay Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Gizmo Student Exploration Nuclear Decay Answer Key
 - Highlighting and Note-Taking Gizmo Student Exploration Nuclear Decay Answer Key
 - Interactive Elements Gizmo Student Exploration Nuclear Decay Answer Key
8. Staying Engaged with Gizmo Student Exploration Nuclear Decay Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Gizmo Student Exploration Nuclear Decay Answer Key
9. Balancing eBooks and Physical Books Gizmo Student Exploration Nuclear Decay Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Gizmo Student Exploration Nuclear Decay Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Gizmo Student Exploration Nuclear Decay Answer Key
 - Setting Reading Goals Gizmo Student Exploration Nuclear Decay Answer Key
 - Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Gizmo Student Exploration Nuclear Decay Answer Key
 - Fact-Checking eBook Content of Gizmo Student Exploration Nuclear Decay Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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