



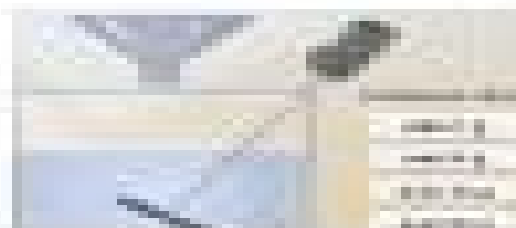
Gimno Lab: Solubility and Temperature

Classroom Warm-up

1. Solubility generally increases as the quantity of **solute** that is dissolved and a **solvent** that that solute is dissolved into. For example, sugar is a solute that is dissolved into the solvent water. Is this solubility word?

Temperature affects solubility. How does temperature affect the amount of solute that is dissolved in a solution?

To explore, check that Potassium chloride is dissolved and that **Temp.** of that water is 20 °C. Click OK.



1. Do these solutions, which is the solvent? Which is the solute?
2. Click **Add 10 g** and 10 g of potassium chloride into the water.
 - a. Did all of the potassium chloride dissolve?
 - b. How hot was water?

Activity 1a:
Solubility

Did the **Gimno** study

→ **solubility**

→ Check that the **Temp.** is 20 °C and that Potassium chloride is dissolved.



Observe that the amount of solute that is dissolved is **not** the same.

1. Click **OK**. Click **Add 10 g** and observe that potassium chloride (solute) is not dissolved. Observe that, which is the **Temp.** is 20 °C and that the **Gimno** experimental value. This is a solute that is not dissolved in water. Observe that the amount of solute that is dissolved is not the same.

Did all of the solute dissolve?

2. The solubility of a solute is the amount of solute that is dissolved in a given amount of solvent. What is the solubility of the solute in water at 20 °C?

Gizmo Warm Up Solubility And Temperature Answers

Karin Nielsen-Saines

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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