

Can You Find the Mole Ratio?

In this experiment, baking soda (NaHCO_3) reacts with hydrochloric acid (HCl) to produce sodium chloride, carbon dioxide, and water. You will determine the moles of reactant used and moles of product produced using both the lab data and stoichiometry and compare the two.

Data Table:

mass of test tube and baking soda	19.46 g
mass of empty test tube	19.21 g
mass of baking soda	0.25 g
mass of test tube and sodium chloride	19.39 g
mass of empty test tube	19.21 g
mass of sodium chloride	0.18 g

Conclusion Questions:

1. Calculate the number of moles of baking soda used in the lab.

$$0.25 \text{ g NaHCO}_3 \left(\frac{\text{mole}}{84.01 \text{ g}} \right) = 0.00297 \text{ moles of NaHCO}_3$$

2. Calculate the number of moles of sodium chloride produced in the lab.

$$0.18 \text{ g of NaCl} \left(\frac{\text{mole}}{58.44} \right) = 0.0031 \text{ moles of NaCl}$$

3. What is the experimental mole ratio of baking soda (NaHCO_3) to sodium chloride (NaCl)?

$$1:1$$

4. Write a balanced equation for the reaction that took place in the experiment.



5. According to the balanced equation, what is the theoretical mole ratio of baking soda to sodium chloride?

$$1:1$$

6. Was the experimental mole ratio exactly the same as the theoretical mole ratio? If not, give some possible reasons.

Yes ~ if it wasn't some reasons

side reactions
measurement

transfer
contamination

Lab Stoichiometry Datasheet Key

KJ Lindholm-Leary



Lab Stoichiometry Datasheet Key:

This book delves into Lab Stoichiometry Datasheet Key. Lab Stoichiometry Datasheet Key is a crucial topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Lab Stoichiometry Datasheet Key, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Lab Stoichiometry Datasheet Key
- Chapter 2: Essential Elements of Lab Stoichiometry Datasheet Key
- Chapter 3: Lab Stoichiometry Datasheet Key in Everyday Life
- Chapter 4: Lab Stoichiometry Datasheet Key in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Lab Stoichiometry Datasheet Key. The first chapter will explore what Lab Stoichiometry Datasheet Key is, why Lab Stoichiometry Datasheet Key is vital, and how to effectively learn about Lab Stoichiometry Datasheet Key.
3. In chapter 2, the author will delve into the foundational concepts of Lab Stoichiometry Datasheet Key. This chapter will elucidate the essential principles that need to be understood to grasp Lab Stoichiometry Datasheet Key in its entirety.
4. In chapter 3, the author will examine the practical applications of Lab Stoichiometry Datasheet Key in daily life. This chapter will showcase real-world examples of how Lab Stoichiometry Datasheet Key can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Lab Stoichiometry Datasheet Key in specific contexts. This chapter will explore how Lab Stoichiometry Datasheet Key is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Lab Stoichiometry Datasheet Key. This chapter will summarize the key points that have been discussed throughout the book.

The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Lab Stoichiometry Datasheet Key.

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