

Sulfur clock lab

Experimental procedure

1. Measurement of the formation of visible product
2. Measurement of the initial concentration of reactants
3. Prepare solutions A, B, C, D and E as shown:

see Table 10.10.10.10

	Reactants	Reagents	Apparatus
A	1 mL	0.1 M H_2O_2 , 0.01 M KSCN	Reaction setup
B	1 mL	0.1 M H_2O_2 , 0.01 M KSCN	Reaction setup
C	1 mL	0.1 M H_2O_2 , 0.01 M KSCN	Reaction setup
D	1 mL	0.1 M H_2O_2 , 0.01 M KSCN	Reaction setup

4. Add the solution using a graduated cylinder with completely standard glassware. Record the reading.
5. Transfer exactly the volume of 0.1 M H_2O_2 to the reaction setup using a graduated cylinder. Record the volume.
6. Add exactly 0.01 M KSCN to the reaction setup using a graduated cylinder.
7. Prepare and record the data for the reaction.
8. Prepare solutions A, B, C, D and E as shown:
 - a. Transfer 0.1 M H_2O_2 to the reaction setup using a graduated cylinder.
 - b. Add 0.01 M KSCN to the reaction setup using a graduated cylinder.
 - c. Add the 0.1 M H_2O_2 to the reaction setup using a graduated cylinder.
 - d. Add the 0.01 M KSCN to the reaction setup using a graduated cylinder.
9. Transfer solution A to the reaction setup using a graduated cylinder. Do not stir the solution with any efficient stirrer.

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