

Experiment 1

1. Write the general equation for an equilibrium/phase transition and give the conditions of each. Example: melting

Experiment 2: Liquid and Vapor Equilibrium in a Closed System

In a closed system, equilibrium between liquid and vapor is reached. a. Write the general equation for the transition. b. Give the conditions of the transition. c. Give the range of the transition.

- a. liquid $\rightarrow$ vapor b. $T < T_c$, $P < P_c$

a.	b.
c.	d.

Experiment 3: Vapor Equilibrium in a Closed System

1. Write the general equation for the transition between liquid and vapor. a. Give the conditions of the transition. b. Give the range of the transition.

- a. liquid $\rightarrow$ vapor b. $T < T_c$, $P < P_c$

c. $T < T_c$, $P < P_c$

- d. $T < T_c$, $P < P_c$

- e. $T < T_c$, $P < P_c$

- f. $T < T_c$, $P < P_c$

2. Write the general equation for the transition between liquid and vapor. a. Give the conditions of the transition. b. Give the range of the transition.

- a. $T < T_c$

- b. $T < T_c$

How To Find Answers On Lesson Masters

VM Jensen



How To Find Answers On Lesson Masters:

Reviewing **How To Find Answers On Lesson Masters**: Unlocking the Spellbinding Force of Linguistics

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