

The Half-Life of Reactions

Zero-Order

$$t_{1/2} = \frac{[A]_0}{2k}$$

Larger $[A]_0$

Larger $t_{1/2}$

$[A]_0$   $t_{1/2}$

First-Order

$$t_{1/2} = \frac{0.693}{k}$$

constant

$t_{1/2}$ does not
depend on $[A]_0$

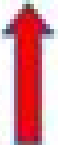
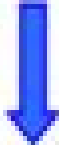
$[A]_0$  $t_{1/2}$

Second-Order

$$t_{1/2} = \frac{1}{k[A]_0}$$

Larger $[A]_0$

Smaller $t_{1/2}$

$[A]_0$   $t_{1/2}$

Half Life Chem Packet

Tao Wei



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