

Griddable Science Items

Topography map

Density

Atoms

$$F = ma$$

			6	.	2	5
0	0	0	0		0	0
1	1	1	1		1	1
2	2	2	2		2	2
3	3	3	3		3	3
4	4	4	4		4	4
5	5	5	5		5	5
6	6	6	6		6	6
7	7	7	7		7	7
8	8	8	8		8	8
9	9	9	9		9	9

Designed for

8th grade STAAR Test



but would work for other standardized tests

Crackables

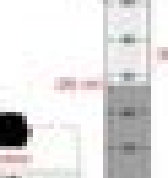
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1. A balance and a graduated cylinder are used to determine the density of a mineral sample. The sample has a mass of 75.1 g and a volume of 14.2 cm³. What is the density of the mineral sample in g/cm³?

$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{75.1}{14.2} = 5.3 \text{ g/cm}^3$$

2. 1 mL = 1 cm³

What is the density of the metal in g/cm³?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{100}{20} = 5.0 \text{ g/cm}^3$$

3. Several glass containers about 10 cm in height are shown. Be sure to use the correct units.

1.



2.



3.



4. What is the density of the liquid in the graduated cylinder?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{100}{20} = 5.0 \text{ g/cm}^3$$

5. The sample of metal has a mass of 80 g. What is the density in g/cm³?



$$\text{density} = \frac{\text{mass}}{\text{volume}} = \frac{80}{125} = 0.64 \text{ g/cm}^3$$

Design

81

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