

Effects and Causes of Global Warming and Climate Change

We are all told the effects of the dooming, inevitable climate change. The first images sparked in our minds are of a dried up, dusty ocean and a sad polar bear swimming. And when we think of these extreme examples we can easily find solace in seeing our actual world and its seemingly mild changes. The thought, "Well it's not that bad" comes to mind when looking out your window. However, if you pay more attention, the very real and rapidly worsening changes in our world are astounding. These changes affect all aspects and categories of our environment: land, sea, and air. Ocean Acidification, Drought, and Natural Disasters are all prominent examples of the diversity of these changes, each caused by obscenely high carbon dioxide levels.

Known as "Climate Change's equally evil twin," The Ocean Portal Team, Ocean Acidification is causing carnage to marine life. Ocean Acidification is the continual decrease in the ocean's pH levels, which is caused by the carbon dioxide from the air dissolving into the seawater. The carbonic acid is primarily stored in the deepest parts of the ocean, not really affecting the shallower levels, where most animals live. So where is the problem? Upwelling is a natural current that raises the deep water to the surface. This occurs when the winds along the coast, push the shallow levels away, therefore causing the deep water to replace the shallow. So, when this upwelling happens, the carbonic acid comes to the surface, where it wrecks havoc on the ecosystem. What is important to remember is that, according to NOAA Pacific Marine Environmental Laboratory, the ocean is absorbing 1/3rd of the world's carbon dioxide-which is roughly 22 million tons a day.

The ocean does have a process in which it can naturally balance its pH levels, called Buffering. However, there is too much carbon dioxide being absorbed, and as a result, the ocean cannot keep up in rebalancing itself. Because of this imbalance, marine life is suffering. Due to the pH levels lowering, the water is becoming more and more acidic, causing shell erosion of clams, oysters, urchins, and pteropods. These protective shells are mainly made of calcium carbonate, a normally insoluble compound. However, in the increasingly acidic water, the shells are becoming more and more soluble, causing these now unprotected animals to die in record numbers. One of these animals being Pteropods, who might seem a bit unimportant. However, because of their place on the food chain being at the very bottom, they are therefore the foundation of all other animals food supply. The Pteropods are just one example out of countless others.

A common misconception from people regarding animal habitats being destroyed, is, "Well, animals can adapt." Yes, they can. However, they are unable to adapt as fast as their environments are changing. It is crucial to remember that these problems are not getting fixed, they are getting worse by the day, and are b

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