

WYŚWIETLACZ CYFROWY

EKRAN STANDARDOWY rys. 12

Na ekranie standardowym wyświetlane są następujące wskazania:

- A** Pozycja korektora świateł reflektorów (tylko przy włączonych światłach mijania)
- B** Zegar (zawsze wyświetlany, również kiedy wyjęty jest kluczyk i zamknięte są drzwi przednie)
- C** Gear Shift Indication (wskazanie zmiany biegu)
- D** Licznik kilometrów (wyświetla kilometry, lub mile, przebiegu)
- E** Wskazanie funkcji Auto-Start-Stop

Uwaga Kiedy kluczyk jest wyjęty (w momencie otwarcia przynajmniej jednego z drzwi przednich) następuje podświetlenie wyświetlacza, który przez kilka sekund wskazuje godzinę i przebieg w kilometrach lub w milach.

GEAR SHIFT INDICATION



Wskazanie zmiany biegu umożliwia, w samochodach z mechaniczną skrzynią biegów, proponowanie kierowcy wykonania zmiany biegu (przełożenie wyższe: shift up lub przełożenie niższe: shift down), co następuje za pośrednictwem odpowiedniego wskazania w zestawie wskaźników.



rys. 12

Sugerowanie zmiany biegu zostało wprowadzone w celu zoptymalizowania zużycia paliwa i stylu jazdy.

Uwaga Wskazanie w zestawie wskaźników pozostaje włączone do momentu wykonania przez kierowcę zmiany biegu lub do momentu, w którym warunki jazdy pozostawiają być na poziomie niekoniecznie wymagającym zmiany biegu w celu zoptymalizowania zużycia paliwa.

PRZYCISKI STERUJĄCE rys. 13

- +** Do przewijania na ekranie odpowiednich opcji w górę lub w celu zwiększenia wyświetlanej wartości.

MENU Naciśnij krótko, aby wejść do menu i/lub przejść do następnego ekranu lub potwierdzić żądany wybór.
ESC Naciśnij długo, aby powrócić do ekranu standardowego.



rys. 13

- Do przewijania na ekranie odpowiednich opcji w dół lub w celu zmniejszenia wyświetlanej wartości.

Uwaga Przyciski + i - służą do aktywacji różnych funkcji w zależności od następujących sytuacji:

Menu ustawień

- w obrębie tego menu umożliwiają przewijanie pozycji w górę lub w dół;
- podczas operacji ustawień umożliwiają zwiększenie lub zmniejszenie wartości.

POTWIERDZENIE SAMOCHODU	
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Ford Ka Mkuser Manual

JA Banks



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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g}) \quad \Delta h = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g}) \quad \Delta h = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{g}) \quad \Delta h = 283 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g})$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 2 \text{ H}_2\text{O}(\text{l}) \quad \Delta h = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{ O}_2(\text{g}) \rightarrow 2 \text{ CO}_2(\text{g}) + 3 \text{ H}_2\text{O}(\text{l}) \quad \Delta h = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{ O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l}) \quad \Delta h = 285 \text{ kJ/mole}$

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web mar 13 2023 be able to use hess s law to calculate reaction enthalpies first law of thermodynamics energy is the ability to do work or transfer heat work is the transfer of energy from one body to another in a sense work is energy in the process of transfer

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values to determine the heat of reaction for the reaction c_2h_4

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web thus for reaction 9 2 in which 1 mole of h_2o and 1 mole of oh^- combine to form 2 moles of h_2o $\Delta H = -55.8 \text{ kJ}$ as shown note that the enthalpy change $\Delta H_{\text{products}} - \Delta H_{\text{reactants}}$ is positive if heat is absorbed that is if $\Delta H_{\text{products}} > \Delta H_{\text{reactants}}$

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