



Quantitative Chemistry

Multiple Choice Questions

Set 1

Tick **one** box.

1. What is the correctly balanced symbol equation for the reaction between magnesium and hydrochloric acid?



2. What is the relative formula mass of dibromoethane ($\text{C}_2\text{H}_4\text{Br}_2$)?
Relative atomic masses (A_r): C = 12, H = 1, Br = 80

A. 93 ☐

B. 186 ☐

C. 188 ☐

D. 372 ☐

3. Which statement about moles is **not** true?

A. all substances contain the same number of moles ☐

B. one mole of carbon contains 6.02×10^{23} atoms of carbon ☐

C. the mass of one mole of a substance in grams is numerically equal to the relative formula mass of that substance ☐

D. the number of atoms, molecules or ions in one mole of a substance is the Avogadro constant ☐

4. What is the mass of two moles of rubidium?

Relative atomic mass (A_r) = 85

A. 42.5g ☐

B. 85g ☐

C. 127.5g ☐

D. 170g ☐

5. How many moles are there in 22g of carbon dioxide (CO_2)?

Relative atomic masses (A_r): C = 12, O = 16

A. 0.5 ☐

B. 2 ☐

C. 12 ☐

D. 968 ☐

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Lei Shi



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