HR•Assessments *SkillSeries*™

# Math Skills

Developed By J. M. Llabre, Ph.D.

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SCORE

Below are 20 problems that measure your ability to perform basic mathematical calculations. You are to place a **checkmark** in front of the most appropriate response. A sample problem similar to those found in the test is presented to the right.

You will have **8 minutes** to answer as many problems as you can.

**Sample Problem:**

3852
+ 97

☐ 3779
☑ 3949
☐ 3910
☐ 4049

**STOP. DO NOT BEGIN UNTIL YOU ARE INSTRUCTED TO DO SO.**

1. $43.75 + 2.50
☐ $45.25 ☐ $46.25
☐ $45.35 ☐ $46.35

2. 17.53 + 3.25
☐ 13.27 ☐ 14.27
☐ 13.28 ☐ 14.28

3. $7.15 × 3
☐ $20.45 ☐ $21.35
☐ $20.55 ☐ $21.45

4. Forty pennies is equivalent to how many nickels?
☐ 6 ☐ 8 ☐ 10 ☐ 12

5. $100 plus 6% sales tax is equivalent to:
☐ $106 ☐ $108 ☐ $112 ☐ $160

6. 216.02 + 72.09
☐ 288.01 ☐ 298.01
☐ 288.11 ☐ 298.11

7. $4,537 − 2,642
☐ $1,875 ☐ $1,895
☐ $1,885 ☐ $2,895

8. If the office furniture costs $9,000 and it was three times as expensive as the office computer, what was the cost of the computer?
☐ $1,450 ☐ $1,800 ☐ $2,300 ☐ $3,300

9. 4200 ÷ 6 =
☐ 600 ☐ 800 ☐ 700 ☐ 900

10. $5,862.00
4,750.00
342.00
+ 1,027.00
☐ $11,981.00 ☐ $12,981.00
☐ $11,991.00 ☐ $12,991.00

11. 10% of $640.00 =
☐ $5.40 ☐ $27.00 ☐ $64.00 ☐ $100.00

12. Five dollars is equivalent to how many dimes?
☐ 25 ☐ 30 ☐ 40 ☐ 50

13. 7,543,025 − 250,570
☐ 7,292,655 ☐ 7,392,655
☐ 7,292,755 ☐ 7,392,755

14. If a company tripled its sales in three years and sales were $1,500,000 the first year, what were the company's sales the third year?
☐ $3,000,000 ☐ $3,500,000 ☐ $4,500,000 ☐ $5,000,000

15. 90,500
325,400
613,880
+ 78,770
☐ 1,100,800 ☐ 1,102,800
☐ 1,101,500 ☐ 1,103,800

16. Seven dollars is equivalent to how many quarters?
☐ 24 ☐ 28 ☐ 32 ☐ 34

17. $3,500,750 − 2,375,425
☐ $1,124,225 ☐ $1,125,225
☐ $1,124,325 ☐ $1,125,325

18. If a company has 400 employees and is projected to increase its employee population to 500 by the end of the year, what percentage increase is projected?
☐ 5% ☐ 10% ☐ 25% ☐ 30%

19. 30% of $1,200 =
☐ 300 ☐ 420 ☐ 360 ☐ 450

20. If 250 packages were delivered in 25 days, and provided the same number of packages were delivered each day, how many packages were delivered each day?
☐ 10 ☐ 15 ☐ 20 ☐ 25

# Job Interview Math Practice Test

Judd E. Hollander

**Job Interview Math Practice Test:**

**Job Interview Math Practice Test** Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Job Interview Math Practice Test**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at agiven pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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boiling point determination lab report conclusion objectivesto correctly measure boiling issues of various liquids and identify an unknown liquid based on its boiling point background information objekte can be distinguished into acid properties and physical properties

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affect by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound s normal boiling point refers to its boiling point at a pressure of 760 text mm ce hg

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine the vapour pressure of a pure liquid at various temperatures

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds tocompare by following the trendline of the straight chain alkanes it could be concluded that asthe molar mass of the straight chain alkanes increases their boiling points increase as well

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should be based on observations made during the experiment

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laboratory partners samantha and mackenzie goals  correctly determine the boiling point of an unknown liquid  measure the refractive index of an unknown  use the gas chromatogram to determine the mass percentage of a mixture and

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50 c so in order to convert an experimental measurement taken in calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that need to be determined by physical properties such as boiling point melting point density solubility etc in this experiment i am to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a bunsen burner boiling

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there are a variety of methods by which a sample s boiling point can be determined including distillation reflux and by using a thiele tube the most straightforward method uses a thiele tube and has the advantage of using less than 0 5ml 0 5 ml of material

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