

F321: Atoms, Bonds and Groups

Structure & Bonding

1. This question is about different models of bonding and molecular shapes.

Magnesium sulfide shows ionic bonding.

- (i) What is meant by the term *ionic bonding*?

.....
.....

[1]

- (ii) Draw a 'dot-and-cross' diagram to show the bonding in magnesium sulfide. Show outer electron shells only.

[2]

[Total 3 marks]

2. 'Dot-and-cross' diagrams can be used to predict the shape of covalent molecules.

Fluorine has a covalent oxide called difluorine oxide, F_2O . The oxygen atom is covalently bonded to each fluorine atom.

- (i) Draw a 'dot-and-cross' diagram of a molecule of F_2O .
Show outer electron shells only.

[2]

Friday 23rd May 2014 Atoms Bonds And Groups

Alan Holden



Friday 23rd May 2014 Atoms Bonds And Groups:

The Annalist, 1932 **The Compact Edition of the Oxford English Dictionary** Sir James Augustus Henry Murray, 1971 Micrographic reproduction of the 13 volume Oxford English dictionary published in 1933 Bonds Between Atoms Alan Holden, 1971 **Atoms, Chemical Bonds, and Bond Dissociation Energies** Sándor Fliszár, 1994 Chemical bonds their intrinsic energies in ground state molecules and the energies required for their actual cleavage are the subject of this book The theory modelled after a description of valence electrons in isolated atoms explains how intrinsic bond energies depend on the amount of electronic charge carried by the bond forming atoms It also explains how bond dissociation depends on these charges While this theory vividly explains thermochemical stability future research could benefit from a better understanding of bond dissociation if we learn how the environment of a molecule affects its charges we also learn how it modifies bond dissociation in that molecule This essay is aimed at theoretical and physical organic chemists who are looking for new perspectives to old problems Atoms and Molecules Roger Harrison, 1972

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