d scattering is incorrect:

vely charged object. increase in the impact parameter . oulomb interaction with the outer valence 4-d

es are deflected through 45° or more by a 0.1 µm gold

A MILLIA ISLAMIA

obtained in the Haemission spectrum from hydrogen

λ) of an atomic spectral line of wavelength 500 nm

correctly describes a system of two identical, nonand 'b' ?

$$\psi_A = 1/2[\psi_a(1)\psi_b(2) - \psi_a(2)\psi_b(1)]$$

= $[\psi_a(1)\psi_b(2) - \psi_a(2)\psi_b(1)]$
: $\psi_A = 1/\sqrt{2}[\psi_a(1)\psi_b(2) - \psi_a(2)\psi_b(1)]$
: $\psi_A = 1/\sqrt{2}[\psi_a(1)\psi_b(2) + \psi_a(2)\psi_b(1)]$

described by the term symbol 32S12. What are the valence electron?

- 65. Which of the following statements
 - (A) Molecular rotational states are
 - (B) The selection rule for molecul-
 - (C) Hydrogen molecule (H₂) does
 - (D) Transitions between low lying



- A 50 kW broadcasting antenna em photons are emitted per second are
 - $(A)1.5 \times 10^{31}$
 - (B) 1.7 × 1031
 - $(C)1.9 \times 10^{31}$
 - $(D)2 \times 10^{31}$
- 67. A photon is incident on an electron observed to be 5 times the wavelength the wavelength of the incident photon

$$(A)2 \times 10^{-13} \text{m}$$

$$(B)3 \times 10^{-13} m$$

$$(C)4 \times 10^{-13} m$$

$$(D)5 \times 10^{-13} m$$

68. The de Broglie wavelength of a par box of length L. is

- (D)L/2
- 69. The operator $\left(\frac{d}{dx} + x\right)\left(\frac{d}{dx} x\right)$ is

$$(A)\frac{d^2}{dx^2} - x^2$$

$$(B)\frac{d^2}{dx^2} - x^2 - 1$$

$$(C)\frac{d^3}{dx^2} - 2x\frac{d}{dx} +$$

(A)
$$\frac{d^2}{dx^2} - x^2$$

(B) $\frac{d^2}{dx^2} - x^2 - 1$
(C) $\frac{d^2}{dx^2} - 2x \frac{d}{dx} + 1$
(D) $\frac{d^2}{dx^2} - 2x \frac{d}{dx} - x^2$

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