

Gas Stoichiometry Practice 1.0 - Answer Key**Objective:** - Perform stoichiometry calculations of gases including those at STP using molar volume**Directions:** Solve each gas stoichiometry problem. Show all work. Include units.

1. Calculate the volume of hydrogen gas that will react with exactly 4.6 L of oxygen gas using the equation below. Assume there is no change in temperature or pressure.



$$4.6 \text{ L O}_2 \left(\frac{2 \text{ L H}_2}{1 \text{ L O}_2} \right) = \boxed{9.2 \text{ L H}_2}$$

2. What volume of CO_2 gas will be produced from 17.5 L of oxygen gas reacting with excess ethane?



$$17.5 \text{ L O}_2 \left(\frac{4 \text{ L CO}_2}{7 \text{ L O}_2} \right) = \boxed{10.0 \text{ L CO}_2}$$

3. What volume of hydrogen at STP can be produced from the reaction of 36.5 g of zinc with hydrochloric acid?



$$36.5 \text{ g Zn} \left(\frac{1 \text{ mol Zn}}{65.38 \text{ g Zn}} \right) \left(\frac{1 \text{ mol H}_2}{1 \text{ mol Zn}} \right) \left(\frac{22.4 \text{ L H}_2}{1 \text{ mol H}_2} \right) = \boxed{12.5 \text{ L H}_2}$$

4. How many grams of NaCl can be produced by the reaction of 12.6 L of chlorine at STP with an excess of sodium?



$$12.6 \text{ L Cl}_2 \left(\frac{1 \text{ mol Cl}_2}{22.4 \text{ L Cl}_2} \right) \left(\frac{2 \text{ mol NaCl}}{1 \text{ mol Cl}_2} \right) \left(\frac{58.44 \text{ g NaCl}}{1 \text{ mol NaCl}} \right) = \boxed{65.7 \text{ g NaCl}}$$

5. In order to produce 50.0 liters of ammonia gas (NH_3) at STP, how many grams of both hydrogen and nitrogen would you need to react?



$$50.0 \text{ L NH}_3 \left(\frac{1 \text{ mol NH}_3}{22.4 \text{ L NH}_3} \right) \left(\frac{1 \text{ mol N}_2}{2 \text{ mol NH}_3} \right) \left(\frac{28.02 \text{ g N}_2}{1 \text{ mol N}_2} \right) = \boxed{21.3 \text{ g N}_2}$$

$$50.0 \text{ L NH}_3 \left(\frac{1 \text{ mol NH}_3}{22.4 \text{ L NH}_3} \right) \left(\frac{3 \text{ mol H}_2}{2 \text{ mol NH}_3} \right) \left(\frac{2.02 \text{ g H}_2}{1 \text{ mol H}_2} \right) = \boxed{6.75 \text{ g H}_2}$$

Matching Key Stoichiometry

K. S. Verma



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Physical Chemistry for JEE Advanced: Part 1, 3E (Free Sample) K. S. Verma, 2022-05-19 Physical Chemistry for JEE Advanced Part 1 a Cengage Exam Crack Series product is designed to help aspiring engineers focus on the subject of physical chemistry from two standpoints To develop their caliber aptitude and attitude for the engineering field and profession To strengthen their grasp and understanding of the concepts of the subjects of study and their applicability at the grassroots level Each book in this series approaches the subject in a very conceptual and coherent manner While its illustrative solved examples facilitate easy mastering of the concepts and their applications an array of solved problems exposes the students to a variety of questions that they can expect in the examination The coverage and features of this series of books make it highly useful for all those preparing for JEE Main and Advanced and aspiring to become engineers

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Basics for Chemistry David A. Ucko, 2013-09-24 Basics of Chemistry provides the tools needed in the study of General Chemistry such as problem solving skills calculation methods and the language and basic concepts of chemistry The book is designed to meet the specific needs of underprepared students Concepts are presented only as they are needed and developed from the simple to the complex The text is divided into 18 chapters each covering some particular aspect of chemistry such as matter energy and measurement the properties of atoms description of chemical bonding study of chemical change and nuclear and organic chemistry Undergraduate students will find the book as a very valuable academic material

Encyclopedic Dictionary of Pyrotechnics K.L. Kosanke, Barry T. Sturman, Robert M. Winokur, B.J.

Kosanke, 2012-10 **Frontiers of Thin Film Technology**, 2000-11-07 Frontiers of Thin Film Technology Volume 28

focuses on recent developments in those technologies that are critical to the successful growth fabrication and characterization of newly emerging solid state thin film device architectures Volume 28 is a condensed sampler of the Handbook for use by professional scientists engineers and students involved in the materials design fabrication diagnostics and measurement aspects of these important new devices

Crystallography of Modular Materials Giovanni Ferraris, Emil Makovicky, Stefano Merlino, 2004-03-25 This is the first book to provide a comprehensive treatment of theories and applications in the rapidly expanding field of the crystallography of modular materials Molecules are the natural modules

from which molecular crystalline structures are built Most inorganic structures however are infinite arrays of atoms and some kinds of surrogate modules e g co ordination polyhedra are usually used to describe them In recent years the attention has been focused on complex modules as the basis for a systematic description of polytypes and homologous polysomatic series modular structures This representation is applied to the modelling of unknown structures and understanding nanoscale defects and intergrowths in materials The Order Disorder OD theory is fundamental to developing a systematic theory of polytypism dealing with those structures based on both ordered and disordered stacking of one or more layers Twinning at both unit cell and micro scale together with disorder causes many problems demons for computer based methods of crystal structure determination This book develops the theory of twinning with the inclusion of worked examples converting the demons into useful indicators for unravelling crystal structure In spite of the increasing use of the concepts of modular crystallography for characterising understanding and tailoring technological crystalline materials this is the first book to offer a unified treatment of the results which are spread across many different journals and original papers published over the last twenty years

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Synthesis of Polymers Dieter A. Schlüter, Craig Hawker, Junji Sakamoto, 2012-05-14 Polymers are huge macromolecules composed of repeating structural units While polymer in popular usage suggests plastic the term actually refers to a large class of natural and synthetic materials Due to the extraordinary range of properties accessible polymers have come to play an essential and ubiquitous role in everyday life from plastics and elastomers on the one hand to natural

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