

Chemistry Worksheets Class 11 on Chapter 3 Classification of Elements and Periodicity in Properties - Set 1

Q-1: What happens when a neutral atom is converted into an anion?

- a) Atomic weight increases
- b) Atomic weight decreases
- c) Size increases
- d) Size decreases

Q-2: The element with the highest affinity for electrons will belong to

- a) Period 2, Group 17
- b) Period 2, Group 18
- c) Period 3, Group 17
- d) Period 2, Group 1

Q-3: Which of the following is not a representative element?

- a) $Z = 38$
- b) $Z = 31$
- c) $Z = 54$
- d) $Z = 26$

Q-4: Find the incorrect statement.

- a) For group 2, the valence electron and valency are identical.
- b) Metal, non-metal, and metalloids are all present in P-block elements.
- c) Helium (He) is the only noble gas with two valence electrons.
- d) The smallest element on the periodic table is He.

Q-5: The symbol of an element with atomic number $Z = 109$ is

- a) Unp
- b) Uns
- c) Uno
- d) Une

Q-6: Which among the following will have the largest atomic radii based on their positions in the periodic table?

Be, N, O, Ne

Mastering Chemistry Answers Chapter 10

RM Cervero

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