

Exercise 14.90 with feedback

You may want to reference ([10 pages 535 - 543](#)) section 14.7 while completing this problem.

A 0.10 m aqueous solution of an ionic compound with the formula MX has a freezing point of $-2.3\text{ }^{\circ}\text{C}$.

Part A

Calculate the van't Hoff factor (i) for MX at this concentration.

Express your answer using two significant figures.

$i =$

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Mastering Chemistry Answer Key Chapter 1

Victor M. Corman



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Mastering Chemistry Answer Key Chapter 1 Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the power of words has be more evident than ever. They have the capability to inspire, provoke, and ignite change. Such is the essence of the book **Mastering Chemistry Answer Key Chapter 1**, a literary masterpiece that delves deep to the significance of words and their effect on our lives. Published by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall affect readers.

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