

Exercise 14.90 with feedback

You may want to reference ([10 pages 535 - 543](#)) section 14.7 while completing this problem.

A 0.10 m aqueous solution of an ionic compound with the formula MX has a freezing point of -2.3°C .

Part A

Calculate the van't Hoff factor (i) for MX at this concentration.

Express your answer using two significant figures.

$i =$

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Mastering Chemistry Answer Key Chapter 1

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Table of Contents Mastering Chemistry Answer Key Chapter 1

1. Understanding the eBook Mastering Chemistry Answer Key Chapter 1
 - The Rise of Digital Reading Mastering Chemistry Answer Key Chapter 1
 - Advantages of eBooks Over Traditional Books
2. Identifying Mastering Chemistry Answer Key Chapter 1
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Mastering Chemistry Answer Key Chapter 1
 - User-Friendly Interface
4. Exploring eBook Recommendations from Mastering Chemistry Answer Key Chapter 1
 - Personalized Recommendations
 - Mastering Chemistry Answer Key Chapter 1 User Reviews and Ratings
 - Mastering Chemistry Answer Key Chapter 1 and Bestseller Lists
5. Accessing Mastering Chemistry Answer Key Chapter 1 Free and Paid eBooks

- Mastering Chemistry Answer Key Chapter 1 Public Domain eBooks
- Mastering Chemistry Answer Key Chapter 1 eBook Subscription Services
- Mastering Chemistry Answer Key Chapter 1 Budget-Friendly Options
- 6. Navigating Mastering Chemistry Answer Key Chapter 1 eBook Formats
 - ePub, PDF, MOBI, and More
 - Mastering Chemistry Answer Key Chapter 1 Compatibility with Devices
 - Mastering Chemistry Answer Key Chapter 1 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Mastering Chemistry Answer Key Chapter 1
 - Highlighting and Note-Taking Mastering Chemistry Answer Key Chapter 1
 - Interactive Elements Mastering Chemistry Answer Key Chapter 1
- 8. Staying Engaged with Mastering Chemistry Answer Key Chapter 1
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Mastering Chemistry Answer Key Chapter 1
- 9. Balancing eBooks and Physical Books Mastering Chemistry Answer Key Chapter 1
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Mastering Chemistry Answer Key Chapter 1
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Mastering Chemistry Answer Key Chapter 1
 - Setting Reading Goals Mastering Chemistry Answer Key Chapter 1
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Mastering Chemistry Answer Key Chapter 1
 - Fact-Checking eBook Content of Mastering Chemistry Answer Key Chapter 1
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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