



READING

11 **Remember** as you read the words in bold, think about what they mean. What do you think the words in bold mean? Write them down.

PARTICLE THEORY AND STATES OF MATTER

What is the theory of particles? Understanding the nature of matter and the arrangement of particles is the basis of the particle theory of matter. The particle theory of matter explains the states of matter and the changes between them. It also explains the properties of matter and the changes between them.

1 Particle Theory Particle theory is a model of matter made up of tiny particles. These particles have three main properties: they are very small, they are in constant motion, and they have spaces between them. The arrangement and movement of particles explain the states of matter and the changes between them.

2 States of Matter Matter can exist in three primary states: solid, liquid, and gas. Each state has different properties. Solids have a fixed shape and volume. Liquids have a fixed volume but no fixed shape. Gases have no fixed shape or volume. The particles in each state are arranged differently. In solids, particles are packed closely together. In liquids, they are more spread out. In gases, they are very far apart.

3 Changes of State Matter can change from one state to another. These changes are called changes of state. They are caused by heating or cooling. For example, melting is the change from solid to liquid. Boiling is the change from liquid to gas. Freezing is the change from liquid to solid. Condensation is the change from gas to liquid. Sublimation is the change from solid to gas. Deposition is the change from gas to solid.



10 Complete the diagram to show the arrangement of the particles in each state of matter. Label the diagram with the names of the states of matter.

9 What is the change of state called when a solid becomes a liquid?

8 What property of matter is different between the three states of matter? Give an example of each state of matter.

7 How do you know that matter is made up of particles?

Particle Theory and States of Matter Guided Reading

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Michael Brown



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