

## Chapter 9 Carbon Chemistry

**Section 9.2 Substituted Hydrocarbons****(pages 272–274)**

*This section discusses organic compounds that contain atoms of elements other than carbon and hydrogen. It also explains the relationship between the properties of organic compounds and functional groups.*

**Reading Strategy (page 272)**

**Monitoring Your Understanding** As you read, complete the table by connecting each functional group with the type of compound that contains the functional group. For more information on this Reading Strategy, see the **Reading and Study Skills** in the **Skills and Reference Handbook** at the end of your textbook.

Connecting Functional Groups to Types of Compounds	
Functional Group	Type of Compound
–OH	
–COOH	
–NH <sub>2</sub>	

- Name the two main products when methane and chlorine react.
  - \_\_\_\_\_
  - \_\_\_\_\_
- To which environmental problem have researchers connected halocarbons containing chlorine and fluorine? \_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_
- Describe a substituted hydrocarbon. \_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_
- Is the following sentence true or false? The functional group in a substituted hydrocarbon determines the properties of the compound. \_\_\_\_\_

**Alcohols (page 273)**

- Methanol and ethanol are two examples of a class of organic compounds called \_\_\_\_\_.
- The functional group in an alcohol is represented as –OH and is called a(n) \_\_\_\_\_ group.
- Identify two ways a halocarbon can be produced.
  - \_\_\_\_\_
  - \_\_\_\_\_

# Hydrocarbon Compounds Section Review Answer Key

**Roman Wölfel**



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