

21. In which of the following reactions addition does not take place according to Markovnikov's rule ?
(IIT 2000)
- (a) $\text{CH}_3\text{CH}=\text{CH}_2 + \text{HCl} \xrightarrow{\text{ROOR}}$
(b) $\text{CH}_3\text{CH}=\text{CH}_2 + \text{HBr} \xrightarrow{\text{ROOR}}$
(c) $\text{CH}_2=\text{CH}_2 + \text{HI} \xrightarrow{\text{ROOR}}$
(d) $\text{CH}_3\text{CH}=\text{CHCH}_3 + \text{HBr} \xrightarrow{\text{ROOR}}$
22. Ozonolysis of 2-methyl butene-2 yields : (UPC 2003)
- (a) ketone (b) aldehyde
(c) aldehyde & ketone (d) none
23. C_6H_{12} on ozonolysis gave only propanone. Alkene is :
(AFMC 2003)
- (a) $\text{CH}_3\text{CH}_2\text{CH}=\text{C}(\text{CH}_3)_2$
(b) $(\text{CH}_3)_2\text{C}=\text{CH}-\text{CH}_3$
(c) $(\text{CH}_3)_2\text{C}=\text{C}(\text{CH}_3)_2$
(d) $\text{CH}_3\text{CH}_2\text{CH}=\text{CHCH}_2\text{CH}_3$
24. The reaction,
 $\text{CH}_2=\text{CHCH}_3 + \text{HBr} \longrightarrow \text{CH}_3\text{CHBrCH}_3$ is :
(AIIMS 2003)
- (a) Nucleophilic addition
(b) Electrophilic addition
(c) Electrophilic substitution
(d) Free radical addition
25. Which of the following alkenes is most stable ?
(UPC 2003)
- (a) $\text{R}_2\text{C}=\text{CR}_2$ (b) $\text{R}-\text{CH}=\text{CH}-\text{R}$
(c) $\text{RCH}=\text{CH}_2$ (d) $\text{CH}_2=\text{CH}_2$
26. A molecule of HCl adds on 3-methyl but-1-ene. Product formed is : (IIT 2001)
- (a) $\text{CH}_3\text{CH}(\text{CH}_3)-\text{CHCl}-\text{CH}_3$
(b) $\text{CH}_3-\text{C}(\text{Cl})(\text{CH}_3)-\text{CH}_2-\text{CH}_3$
(c) $\text{CH}_3\text{CH}(\text{CH}_3)-\text{CH}_2-\text{CH}_2\text{Cl}$
(d) $\text{CH}_3\text{CH}(\text{CH}_3)-\text{CH}(\text{Cl})-\text{CH}_2\text{CH}_3$
27. The intermediate during the addition of HCl to propene in presence of peroxide is : (IIT 2002)
- (a) $\text{CH}_3\dot{\text{C}}\text{HCH}_2\text{Cl}$ (b) $\text{CH}_3\dot{\text{C}}\text{HCH}_3$
(c) $\text{CH}_3\text{CH}_2\dot{\text{C}}\text{H}_2$ (d) $\text{CH}_3\text{CH}_2\dot{\text{C}}\text{H}$
28. Which of the following exhibit cis, trans-isomerism ?
(a) $\text{CH}_2\text{Br}-\text{CH}_2\text{Br}$
- (b) $\text{CBr}_2=\text{CH}_2$
(c) $\text{CHBr}=\text{CHBr}$
(d) $\text{CBr}_2=\text{CH}-\text{CH}_3$
29. The compound that dehydrates most readily is :
(CPMT 2001)
- (a) R_2CHOH (b) RCH_2OH
(c) R_3COH (d) $\text{R}-\text{OH}$
30. The addition of propene with HOCl proceeds via the addition of : (IIT 2001)
- (a) H^+ in the first step
(b) Cl^+ in the first step
(c) OH^- in the first step
(d) Cl^+ and OH^- in a single step
31. Which compound is sure to be alkene ?
(AFMC 2003)
- (a) C_2H_4 (b) C_3H_6
(c) C_4H_8 (d) C_5H_{10}
(e) All
32. 1-chlorobutane with alc. KOH gives : (AFMC 2001)
- (a) 1-butene (b) 1-butanol
(c) 2-butene (d) 2-butanol
33. How much ethylene is oxidised by 3 litres of O_2 under identical conditions ? [CBSE (Med.) 2002]
- (a) One mole (b) One litre
(c) 3 litre (d) 0.3 litre
34. Which one of the following alkenes will react fastest with H_2 under catalytic hydrogenation condition ?
(IIT (S) 2000)
- (a) $\text{R}_2\text{C}=\text{CH}_2$ (b) $\text{RHC}=\text{CHR}$
(c) $\text{R}_2\text{C}=\text{CHR}$ (d) $\text{R}_2\text{C}=\text{CR}_2$
35. Which of the following gases gives a red precipitate with ammonical solution of cuprous chloride ?
(DCE 2004)
- (a) C_2H_4 (b) C_2H_6
(c) C_2H_2 (d) $\text{CH}_3\text{CH}_2\text{CH}_3$
36. Which will not give a precipitate with Tollen's reagent?
(DCE 2002)
- (a) Ethyne
(b) 2-pentyne
(c) 1-butyne
(d) 3-methyl-1-butyne
37. Which hydrocarbon reacts with sodium in liquid NH_3 ?
(AIEEE 2008)
- (a) $\text{CH}_3-\text{CH}_2-\text{C}\equiv\text{CH}$
(b) $\text{CH}_3-\text{C}\equiv\text{C}-\text{CH}_3$
(c) $\text{CH}_3-\text{CH}=\text{CH}-\text{CH}_3$
(d) $\text{CH}_3-\text{CH}_2-\text{CH}=\text{CH}_2$
38. Increasing order of volatility of C_2H_6 , C_2H_4 , C_2H_2 and C_6H_6 is :
(BHU 2003)

Mcqs Of Hydrocarbons

Xiang Xie



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use this engaging lesson pack to teach year 5 children to solve problems about 2d representations of 3d models this lesson teaches children to relate 3d objects to 2d nets children examine 3d objects from different viewpoints to identify nets and faces this resource includes a lesson presentation and plan and differentiated worksheets