| | iv. | of the unit cell. | | nsity of 8.966 g. cm ⁻³ . Calculate the volume Avogadro number N_A is 6.022×10^{23} mol ⁻¹ . | |
|------|------------|---|--|--|------|
| | | | L mos and | Avogadro number N _A is 0.022 × 10 mol . | [12] |
| Q.3. | Ans: | Answer any SIX of the following: i. What is the action of the following reagents on ammonia: | | | |
| | | Nessler's reagent Sodium metal | | | |
| | 100 | State the first and second law of electr | rolysis. | | |
| | His. | Draw neat and labelled diagram of Bessemer converter used in the extraction of copper. | | | |
| | iv. | Derive the relation between half-life period and rate constant for first order reaction. | | | |
| | v. | Derive the relation between ΔG° and equilibrium constant (K) for the reaction, $aA+bB \Box = cC+dD$. | | | |
| | wi. | Explain brown ring test with the help of chemical equation. | | | |
| | wiii. | Explain, why do aquatic animals prefe | er to stay at | lower level of water during summer? | |
| | viii. | Distinguish between: Crystalline solids and Amorphous soli | ids. | | |
| Q.4. | | Select and write the most appropriate answer from the alternatives given below each sub-question: | | | |
| | i. | following number of valence electrons | s | to be added to silicon should have the | |
| | | (A) 2 (C) 4 | (B) | 3 5 | |
| | 116 | Number of faradays of electricity required to liberate 12 g of hydrogen is | | | |
| | | (A) 1 | (B) | 8 | |
| | | (C) 12 | (D) | 16 | |
| | iii. | What is molecular formula of oleum? | | | |
| | | (A) H ₂ SO ₃ (C) H ₂ S ₂ O ₃ | | H ₂ SO ₄ H ₂ S ₂ O ₅ | |
| | - 14.00.00 | | | | |
| | iv. | Purification of aluminium by electroly (A) Hoope process | | Hall Process | |
| | | (C) Baeyer process | | Serperek process | |
| | v. | The rate of reaction for certain reaction is expressed as: $\frac{1}{3} \frac{d[A]}{dt} = \frac{1}{2} \frac{d[B]}{dt} = -\frac{d[C]}{dt}$ | | | |
| | | The reaction is | | | |
| | | (A) $3A \longrightarrow 2B + C$ | of section 1 | $2B \longrightarrow 3A + C$ | |
| | | (C) 2B + C → 3A | | 3A + 2B → C | |
| | vi. | A system absorbs 640 J heat and does work of 260 J, the change in internal energy of the system will be | | | |
| | | (A) +380 J (C) +900 J | | - 380 J - 900 J | |
| | viii | Which of the following is 'not' a colli | | | |
| | WILL | (A) Vapour pressure | The second secon | Depression in freezing point | |
| | | (C) Elevation in boiling point | (D) | | |
| | | | 1.0 | | |

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going to take these lines slowly among major poets dickinson is about as far away from easy to read prose as you can get you know that handy stuff known as syntax putting words in an order

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