

STD XII - CHEMISTRY

Time : 3 hrs.]

MARCH - 2013

[Marks : 70

SECTION - I

Q. 1. Select and write the most appropriate answers from the given alternatives for each sub-question :

[7]

- (i) In body centred cubic structure the space occupied is about _____.
 (a) 68% (b) 53% (c) 38% (d) 32%
- (ii) For a gaseous reaction the unit of rate of reaction is _____.
 (a) L atm s^{-1} (b) $\text{atm mol}^{-1} \text{s}^{-1}$ (c) atm s^{-1} (d) mol s
- (iii) Which of the following compounds contains S = O as well as S = S bonds?
 (a) Sulphuric acid (b) Thiosulphuric acid
 (c) Sulphurous acid (d) Thiosulphurous acid
- (iv) Which of the following solutions shows maximum depression in freezing point?
 (a) 0.5 M Li_2SO_4 (b) 1 M NaCl (c) 0.5 M $\text{Al}_2(\text{SO}_4)_3$ (d) 0.5 M BaCl_2
- (v) For a chemical reaction $dS = -0.035 \text{ kJ/K}$ and $dH = -20 \text{ kJ}$. At what temperature does the reaction turn nonspontaneous?
 (a) 5.14 K (b) 57.14 K (c) 571.4 K (d) 5714.0 K
- (vi) The standard e.m.f. of the following cell is 0.463 V

$$\text{Cu}_{(s)} \mid \text{Cu}^{++}_{(aq)} (1\text{M}) \parallel \text{Ag}^{++}_{(aq)} (1\text{M}) \mid \text{Ag}_{(s)}$$
 . What is the standard potential of Cu electrode?
 (a) 1.137 V (b) 0.337 V (c) 0.463 V (d) -0.463 V
- (vii) Fe_2O_3 is reduced to spongy iron near the top of blast furnace by _____.
 (a) H_2 (b) CaO (c) SiO_2 (d) CO

Q. 2. Answer any SIX of the following :

[12]

- (i) Distinguish between crystalline solid and amorphous solid.
- (ii) State Kohlrausch Law and write mathematical expression of molar conductivity of the given solution at infinite dilution.
- (iii) Write cell reactions in lead storage battery during discharge.
- (iv) Draw structures and write geometry of PCl_3 and PCl_5 .
- (v) Prove that $\Delta H = \Delta U + \Delta nRT$. What is the condition under which $\Delta U = \Delta H$?
- (vi) Mention names and formulae of two ores of aluminium.
- (vii) Derive the relationship between relative lowering of vapour pressure and molar mass of nonvolatile solute.
- (viii) What is pseudo first order reaction? Give one example of it.

Q. 3. Answer any THREE of the following :

[9]

- (i) Calculate the mole fraction and molality of HNO_3 in a solution containing 12.2% HNO_3 . (Given - atomic masses : H = 1, N = 14, O = 16)
- (ii) Consider the reaction -



What are the values of the following at the same time ?

- (a) $-\frac{d[\text{I}^{-}]}{dt}$ (b) $-\frac{d[\text{S}_2\text{O}_8^{2-}]}{dt}$ (c) $-\frac{d[\text{I}_3^{-}]}{dt}$

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1. Understanding the eBook Hsc Board Exam Chemistry Question 2013 Mcq
 - The Rise of Digital Reading Hsc Board Exam Chemistry Question 2013 Mcq
 - Advantages of eBooks Over Traditional Books
2. Identifying Hsc Board Exam Chemistry Question 2013 Mcq
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Hsc Board Exam Chemistry Question 2013 Mcq
 - User-Friendly Interface
4. Exploring eBook Recommendations from Hsc Board Exam Chemistry Question 2013 Mcq
 - Personalized Recommendations
 - Hsc Board Exam Chemistry Question 2013 Mcq User Reviews and Ratings
 - Hsc Board Exam Chemistry Question 2013 Mcq and Bestseller Lists
5. Accessing Hsc Board Exam Chemistry Question 2013 Mcq Free and Paid eBooks
 - Hsc Board Exam Chemistry Question 2013 Mcq Public Domain eBooks
 - Hsc Board Exam Chemistry Question 2013 Mcq eBook Subscription Services
 - Hsc Board Exam Chemistry Question 2013 Mcq Budget-Friendly Options

6. Navigating Hsc Board Exam Chemistry Question 2013 Mcq eBook Formats
 - ePub, PDF, MOBI, and More
 - Hsc Board Exam Chemistry Question 2013 Mcq Compatibility with Devices
 - Hsc Board Exam Chemistry Question 2013 Mcq Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Hsc Board Exam Chemistry Question 2013 Mcq
 - Highlighting and Note-Taking Hsc Board Exam Chemistry Question 2013 Mcq
 - Interactive Elements Hsc Board Exam Chemistry Question 2013 Mcq
8. Staying Engaged with Hsc Board Exam Chemistry Question 2013 Mcq
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Hsc Board Exam Chemistry Question 2013 Mcq
9. Balancing eBooks and Physical Books Hsc Board Exam Chemistry Question 2013 Mcq
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Hsc Board Exam Chemistry Question 2013 Mcq
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Hsc Board Exam Chemistry Question 2013 Mcq
 - Setting Reading Goals Hsc Board Exam Chemistry Question 2013 Mcq
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Hsc Board Exam Chemistry Question 2013 Mcq
 - Fact-Checking eBook Content of Hsc Board Exam Chemistry Question 2013 Mcq
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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