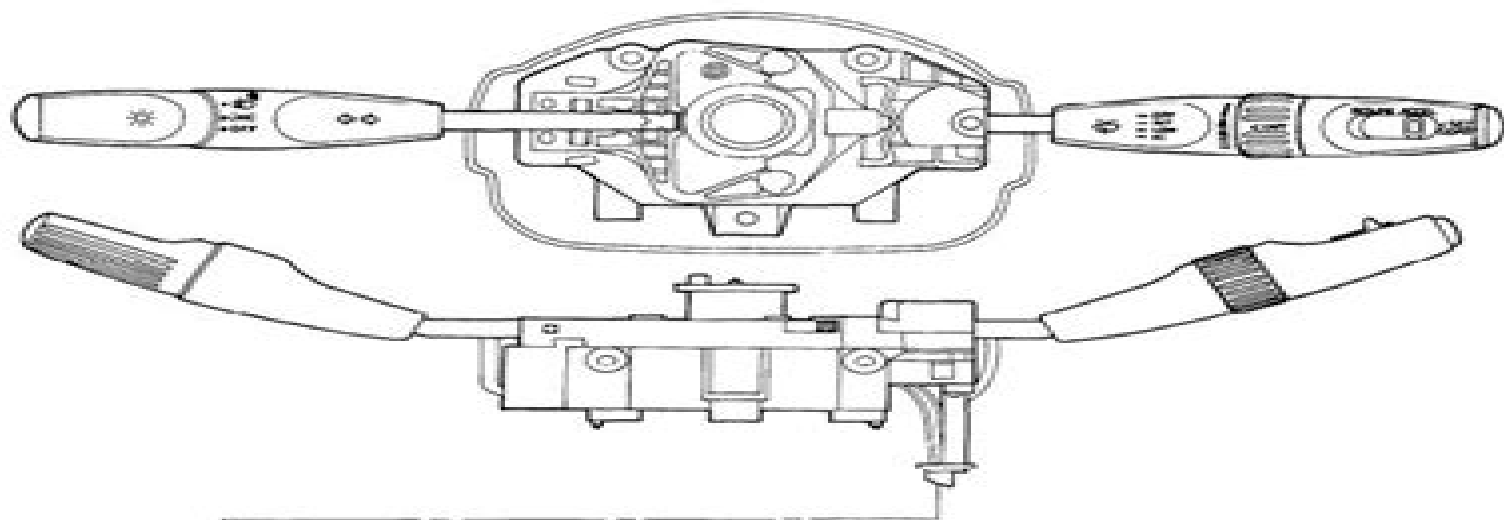
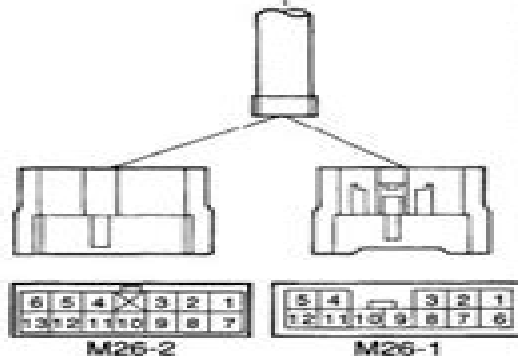


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Circuit connection

Connector No.	Terminal No.	Wire size & color	Circuit	Description
M26-1	1		No Connection	
	2	0.85RL	INT	Intermittent wiper
	3	0.85RW	S	Wiper parking
	4	0.85RY	HI	Wiper high speed
	5	0.85YB	LOW	Wiper low speed
	6		No Connection	
	7	0.5WL	W	Washer switch
	8	0.85GB	HO	Horn
	9	0.85B	EW, EA	Wiper & washer cruise ground
	10	0.3GW	INT(T)	Intermittent wiper
	11	0.3L	SEY	Speed set coast
	12	0.3LgW	RES	Resume accel
M26-2	1	0.3YR	TS	Tail lamp switch
	2	1.25WB	HS(2)	Head lamp switch passing
	3	1.25RG	HU	Head lamp upper beam power
	4	1.25B	HL	Head lamp lower beam power
	5	1.25B	EB	Dimmer & passing switch ground
	6		No Connection	
	7	0.3WL	HS(1)	Head lamp switch
	8	0.3B	EL	Lighting switch ground
	9	0.3G	F	Rear fog lamp
	10	0.5RW	TB	Flasher unit power
	11	0.5RL	TL	Turn signal lamp (LH) switch
	12	0.5RY	TR	Turn signal lamp (RH) switch
	13		No Connection	



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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °C

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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nitrobromobenzene recrystallization was used to purify 4 nitrobromobenzene

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