## Mixed Gas Law Worksheet

Answers available at end

Name:		(show your work)	period:
Convert:			
100 mmHg =	kPa	300 mmHg =	atm
120 atm	mmHg	300 torr =	kPa
25 kPa =	mmHg	25 mmHg =	atm
300°C = K	150K =	°C 5000°C = _	K
300K =°C	-255°C =	K 3500K = _	°C
200°C =°F	95oF =	_°C =K 45°	F =K

 A 70. liter sample of gas initially at 280°C is allowed to cool at constant pressure, what will the new volume be at 0°C?

$$P_1 = P_2 = V_3 = V_3 = T_2 = T_2 = T_3 = T_2 = T_3 = T_4 = T_5 = T_5$$

2) A 4.0 liter sample of gas initially at 25°C is allowed to cool at constant pressure, what will the new volume be at -5°C?

$$P_1 = P_2 = V_3 = V_4 = T_1 = T_2 = T_3 = T_4 = T_5 = T_5$$

3) A buoyancy vest (used in scuba diving) is filled to a volume of 2.3 liters at a pressure of 1 atmosphere (at the surface where the temperature is a balmy 300K), what will the volume of air be in the vest when the diver dives to a depth of about 90 feet where the temperature is 17°C and the pressure is 4 atmospheres?

# **Gas Laws And Scuba Diving Worksheet Answers**

**George I. Sackheim** 

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