#### Gas laws practice test

#### Multiple Choice

Identify the choice that best completes the statement or answers the question.

-	1.	a. volume. c.	length.	
	2.	Why does a can collapse when a vacuum pump removes air from the can?  a. The inside and outside forces balance out and crush the can.  b. The unbalanced outside force from atmospheric pressure crushes the can.  c. The atmosphere exerts pressure on the inside of the can and crushes it.  d. The vacuum pump creates a force that crushes the can.		
	3.	If the height of mercury in a barometer at 0°C is less than 760 mm Hg, then a, the atmospheric pressure is less than standard atmospheric pressure. b. the atmospheric pressure is greater than standard atmospheric pressure. c. the atmospheric pressure is equal to standard atmospheric pressure. d. the atmospheric pressure cannot be determined.		
	4.	Convert the pressure 0.840 atm to mm Hg. a. 365 mm Hg c. b. 437 mm Hg d.	638 mm Hg 780 mm Hg	
	5.	Convert the pressure 1.30 atm to kPa.  a. 2 kPa	132 kPa 245 kPa	
	6.		0°C. 0 K.	
	7.	are all placed in a single 2 L container? a. 247 mm Hg c.	parate 2 L containers. What pressure do they exert if they  1.48 × 10 <sup>3</sup> mm Hg  2.22 × 10 <sup>3</sup> mm Hg	
1	8.	If the temperature remains constant, $V$ and $P$ represent the new volume and pressure, what is the a. $P'V = V'P$ c. b. $VV'' = PP''$ d.		
-	9.	volume does the gas occupy when the pressure do a. 80.0 mL c.	ressure is 800.00 mm Hg. At constant temperature, what creases to 700.0 mm Hg? 600. mL 640. mL	
	10.	a. its volume increases. c.	its volume decreases. its density decreases.	

# **Gas Law Test Questions**

**Annelies Wilder-Smith** 

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