Make and Carlot Co.	ACC Expense of a	Fig. 4. April 1997
Martino .	The state of the s	The Contract of the Contract o

1 Write each of the changes listed below in the correct column in the table to show whether they are physical changes or chemical reactions.

Changes: boiling burning melting neutralising

Physical changes	Chemical reactions	

2 Some of the statements below are true and some are false.

Put a tick (<') in the box next to the statements that are true.

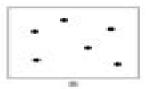
- New substances are made during a chemical reaction.
- New substances are made during a physical change.
- A chemical reaction is usually easier to reverse than a physical change.
- A physical change is usually easier to reverse than a chemical reaction.
- You cannot reverse physical changes or chemical reactions.
- 3 Each observation below is describing a physical change or a chemical reaction.

Put a tick (*') in the correct column to show whether each is a physical change or a chemical reaction.

Observation	Physical change	Chemical reaction
A liquid was set on fire and it burnt with a yellow flame.		
A liquid was cooled down and it changed into a solid.		

4 The boxes A and B show gas particles in containers.





Which box shows the gas at the higher pressure? _

Explain your answer.

I cam....

- Identity and explain the differences between physical changes and chemical reactions
- use particle theory to explain gas pressure and how it can be changed.

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