

Name: Domin: CK As Grede Date:

Student Exploration: Covalent Bonds

Vocabulary: covalent bond, diatomic molecule, Lewis diagram, molecule, noble gases, nonmetal, octet rule, shell, valence, valence electron

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

There are eight markers in a full set, but Flora and Frank each only have seven markers.
Flora is missing the red marker, and Frank is missing the blue marker.

What can they do so that each has a full set of markers? ____Share____

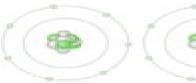
Otto and Olivia each have six markers. Otto is missing the purple and green markers, and Olivia is missing the black and brown markers. What can they do so that each has a full set?

share 2

Gizmo Warm-up

Just like the students described above, **nonmetal** atoms can share electrons. As you will see in the Covalent Bonds Gizmo, atoms form bonds in this way.

To begin, check that Fluorine is selected from the Select a substance menu. Click Play () to see the electrons orbiting the nucleus of each atom.





- The outermost electrons in each atom are called **valence electrons**. How many valence electrons does each fluorine atom have?
- Click Pause (III). Drag an electron from the left atom to the right atom. Click Play.

What happens? it orbits both atoms

3. Click Pause, drag an electron from the right atom to the left, and then click Play.

What happens now? it orb; to both atoms

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Table of Contents Explore Learning Student Exploration Covalent Bonds Answers

- 1. Understanding the eBook Explore Learning Student Exploration Covalent Bonds Answers
 - The Rise of Digital Reading Explore Learning Student Exploration Covalent Bonds Answers
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Explore Learning Student Exploration Covalent Bonds Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Explore Learning Student Exploration Covalent Bonds Answers
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Explore Learning Student Exploration Covalent Bonds Answers
 - Personalized Recommendations
 - Explore Learning Student Exploration Covalent Bonds Answers User Reviews and Ratings
 - Explore Learning Student Exploration Covalent Bonds Answers and Bestseller Lists

- 5. Accessing Explore Learning Student Exploration Covalent Bonds Answers Free and Paid eBooks
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 - Explore Learning Student Exploration Covalent Bonds Answers eBook Subscription Services
 - Explore Learning Student Exploration Covalent Bonds Answers Budget-Friendly Options
- 6. Navigating Explore Learning Student Exploration Covalent Bonds Answers eBook Formats
 - o ePub, PDF, MOBI, and More
 - Explore Learning Student Exploration Covalent Bonds Answers Compatibility with Devices
 - Explore Learning Student Exploration Covalent Bonds Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Explore Learning Student Exploration Covalent Bonds Answers
 - Highlighting and Note-Taking Explore Learning Student Exploration Covalent Bonds Answers
 - Interactive Elements Explore Learning Student Exploration Covalent Bonds Answers
- 8. Staying Engaged with Explore Learning Student Exploration Covalent Bonds Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Explore Learning Student Exploration Covalent Bonds Answers
- 9. Balancing eBooks and Physical Books Explore Learning Student Exploration Covalent Bonds Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Explore Learning Student Exploration Covalent Bonds Answers
- 10. Overcoming Reading Challenges
 - o Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Explore Learning Student Exploration Covalent Bonds Answers
 - Setting Reading Goals Explore Learning Student Exploration Covalent Bonds Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Explore Learning Student Exploration Covalent Bonds Answers
 - Fact-Checking eBook Content of Explore Learning Student Exploration Covalent Bonds Answers
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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