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**APJ ABDUL KALAM TECHNOLOGICAL UNIVERSITY**

First Semester B.Tech Degree Regular and Supplementary Examination December 2020 (2019 scheme)

**Course Code: CYT100****Course Name: ENGINEERING CHEMISTRY  
(2019-Scheme)**

Max. Marks: 100

Duration: 3 Hours

**PART A****Answer all questions, each carries 3 marks.**

- |    |  | Marks |
|----|--|-------|
| 1  | What will be the standard electrode potential of $\text{Ni}^{2+} / \text{Ni}$ electrode if the cell potential of the cell $\text{Ni} / \text{Ni}^{2+}(1\text{M}) // \text{Cu}^{2+}(0.1\text{M}) / \text{Cu}$ is 0.59 V at 25 °C?<br>$E^\circ_{\text{Cu}^{2+}/\text{Cu}} = 0.34 \text{ V}$  | (3)   |
| 2  | Briefly explain the principle of electroless plating.  | (3)   |
| 3  | Give the mechanism of interaction of electromagnetic radiation with oscillating dipole.  | (3)   |
| 4  | State Beer-Lambert's law and write the differential form.  | (3)   |
| 5  | Write any three differences of TGA and DTA.  | (3)   |
| 6  | Explain sol-gel method for the synthesis of nano particles   | (3)   |
| 7  | Determine the configuration of the following alkenes as E or Z:  | (3)   |
|    | <div style="display: flex; justify-content: space-around; align-items: center;"> <div style="text-align: center;"> <math display="block">\begin{array}{c} \text{H}_3\text{C} \quad \quad \text{CH}_2\text{OH} \\ \diagdown \quad \diagup \\ \text{C} = \text{C} \\ \diagup \quad \diagdown \\ \text{H} \quad \quad \text{Cl} \end{array}</math> <p><b>A</b></p> </div> <div style="text-align: center;"> <math display="block">\begin{array}{c} \text{C}_6\text{H}_5 \quad \quad \text{COOH} \\ \diagdown \quad \diagup \\ \text{C} = \text{C} \\ \diagup \quad \diagdown \\ \text{H} \quad \quad \text{CH}_3 \end{array}</math> <p><b>B</b></p> </div> </div> |       |
| 8  | Mention any three advantages of OLEDs over LED and LCD.  | (3)   |
| 9  | Explain disinfection by chlorination.  | (3)   |
| 10 | Compare BOD and COD.   | (3)   |

**PART B****Answer one full question from each module, each question carries 14 marks****Module-I**

- |    |  |      |
|----|--|------|
| 11 | a) Explain the mechanism of electrochemical corrosion in different environmental conditions. | (10) |
|    | b) How is the cell constant of a conductivity cell measured?                                 | (4)  |

# Engineering 2course Chemistry May 2question Paper

**Scott C. Dulebohn**



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