# চট্টগ্রাম বোর্ড

নভাৱ পাত্ৰ FeSO4 প্ৰবণ ৰাখলে সৃষ্ট কোনেৰ কোৰ ১২। [Fe(H<sub>2</sub>O)<sub>6</sub>]<sup>3+</sup> এবিভের অনুবন্ধী ভারক হলো-जिल्ला स्टब (\*) [Fe(H2O)4(OH)-]\*  $[E^*Zn^{2+}(sq)/Zn(s) = -0.76 \text{ V};$ [Fe(H<sub>2</sub>O)<sub>5</sub>(OH)]<sup>2+</sup>  $E^{\circ}_{Fe^{2+}(nq)/Fe(s)} = -0.44 \text{ V}$ (\*f) [Fe(H<sub>2</sub>O)<sub>e</sub>(OH)]<sup>3+</sup> (マ) - 0-32 V (≺) -1-20 V (4) · [Fe(H<sub>2</sub>O)<sub>6</sub> (OH)]<sup>2+</sup> (N) + 1-20 V + 0-32 V = ১৩। তড়িৎ বানায়দিক কোকের emf = ? কোন প্যাপতি ব্যয়েগের সূত্র হতে সবচেয়ে কম বিচ্চাত হবে? 3-7m II (4) CO2 E Anode (ex) + E Cathode (Red) (4) O<sub>2</sub> (40 N<sub>2</sub> E Cathode (Red) - E Anode (Red) হেটারো আরোমেটিক বৌগ কোনটিঃ E Anode (ox) - E Cathode (ox) 400 1 (ক) ইথিলিন অক্সাইড (খ) খেনজিন (ঘ) টেট্রাহাইজ্রোকিউরান কোনটি স্টেক্ (中) 1941 टकान ट्वीएन 'C' এর জারণ মান শূন্য? (M) HI OH 1. 11 10 101 (4) CO (\*) CH<sub>4</sub> 58। 27°C ভাশমানার 8 g CH<sub>4</sub> গ্যালের গতিপত্তি কত যুগা (V) C2H2 CH<sub>2</sub>Cl<sub>2</sub> একই মাত্ৰার বিদ্যুৎ একই সময় চালনা করলে কোন (字) 935-32 1870-65 (41) 3741-30 बारमधि कार्रदारक व्यन्ति कमा दरवा 4870-30 (4) Fe3+ (\*) Cu2+ উদ্দীপকটি লক্ষ কর এবং ১৫ খ ১৬ নং গ্রাহের উত্তর দাও। (20 Na+ SATP OF HIGHT WHITE MICHA WAS  $eL^{-1}$ লেখি খোলাৰ বিকারীয় এবিভ Na<sub>2</sub>CO<sub>3</sub> 50 mL, 0-2 M 1-77 (4) 1-96 STATE OF SEP-PER (P) 1-85 (9) 1·21 উদীপকটি দক্ষ্য কর এবং ২৪ ও ২৫ সং প্রপ্রের উত্তর দাও : THE A PERSONAL PROPERTY. R - CH = CH<sub>2</sub> + HBr H<sub>2</sub> O<sub>2</sub> 'A' (spept (beetly) + B লাত B তার প্রবাণ থাবা A পাতের প্রবণ সম্পূর্ণকলে প্ৰশমিত হয়। ASI A CHIMIP ROTE -(\*) R - CH(OH) - CH<sub>1</sub> 30 I A लाउत्तव अवरणंत चनमाता ppm जन्दन कवा (4) R - CH<sub>2</sub> - CH<sub>2</sub> - OH (47)  $1-0.6 \times 10^3$  $5.3 \times 10^4$ R + CH2 - CH2 Br (№) 1-96 × 10<sup>4</sup>  $9.8 \times 10^{3}$ (W) R - CH(Br) - CH, A পাত্রের প্রবর্গের আয়তন কত? 3/9x 1 Ger Reflerence (4) 40 mL यावकनिकड मीडि श्रद्याका नव 20 mL (N) 2000 mL উৎপাদ A আলোক সভিদা 1000 mL মুক্ত মুলক সৃত্তির মাধ্যমে বিক্রিয়া লংঘটিত হয় ১৭। বেনজিনের কার্বন-কার্বন বছন দৃহত্ব কর nm ? কোনটি সাহিত্য 0-139 (\*) (\*) II (P) 0-154 E 10 AKE (N) 0-120 60 NO 655 (4) 0-134 (S) L N S MI

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web the atomic mass of a single atom is simply its total mass and is typically expressed in atomic mass units or amu by definition an atom of carbon with six neutrons carbon 12 has an atomic mass of 12 amu other atoms don t generally have round number atomic masses for reasons that are a little beyond the scope of this article

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positrons c protons electrons and positrons protons neutrons and electrons  $\boldsymbol{d}$