

CH301 Worksheet 1 Answer Key: High School Chemistry Questions.

For each problem, identify the kind of question BEFORE answering it.

1. A 25 kg ball has a kinetic energy of 225 J. What is the speed of the ball?
What kind of question is this? energy
3 m/s
2. How many molecules of water are there in 100 mL of water? (The density of water = 1.000 g/mL)
What kind of question is this? stoichiometry
 3.34×10^{24} molecules
3. Given that the atomic mass of carbon is given as 12.0107 g/mol, what are the approximate relative abundances of the two isotopes ^{12}C and ^{13}C ? *What kind of question is this? isotopes*
98.93% ^{12}C , 1.07% ^{13}C
4. In the electrolysis of $2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2$, how many atom of O in O_2 are made from 4 g of H_2O ?
What kind of question is this? stoichiometry
 1.34×10^{23} molecules
5. What are the most likely ionic forms of (a) sodium and (b) calcium?
What kind of question is this? periodic trends
(a) Na^+ ; (b) Ca^{2+}
6. What is the product of a reaction between metallic lithium and oxygen, O_2 ?
What kind of question is this? oxidation numbers
 Li_2O
7. Name the following species: (a) Cl^- , (b) NO_3^- , (c) C_3H_8 . *What kind of question is this? naming*
(a) chloride; (b) nitrate; (c) propane
8. The combustion of ethanol has the following equation: $\text{C}_2\text{H}_5\text{OH} + 3 \text{O}_2 \rightarrow 2 \text{CO}_2 + 2 \text{H}_2\text{O}$
What mass of CO_2 is formed from 1 kg of ethanol? *What kind of question is this? stoichiometry*
1.913 kg
9. Vitamin K contains 78.95% C, 3.95% H, and 21.05% O by mass. What is its empirical formula?
What kind of question is this? empirical formula calculation
 $\text{C}_{20}\text{H}_{30}\text{O}$
10. Given that vitamin K (see question 9) has molecular weight of 158.15 g/mol, what is its molecular formula? *What kind of question is this? molecular formula calculation*
 $\text{C}_{40}\text{H}_{60}\text{O}_2$

High School Chemistry Test Questions And Answers

Bill Deffebach



High School Chemistry Test Questions And Answers:

This book delves into High School Chemistry Test Questions And Answers. High School Chemistry Test Questions And Answers is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into High School Chemistry Test Questions And Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:
 - Chapter 1: Introduction to High School Chemistry Test Questions And Answers
 - Chapter 2: Essential Elements of High School Chemistry Test Questions And Answers
 - Chapter 3: High School Chemistry Test Questions And Answers in Everyday Life
 - Chapter 4: High School Chemistry Test Questions And Answers in Specific Contexts
 - Chapter 5: Conclusion
 2. In chapter 1, the author will provide an overview of High School Chemistry Test Questions And Answers. This chapter will explore what High School Chemistry Test Questions And Answers is, why High School Chemistry Test Questions And Answers is vital, and how to effectively learn about High School Chemistry Test Questions And Answers.
 3. In chapter 2, the author will delve into the foundational concepts of High School Chemistry Test Questions And Answers. This chapter will elucidate the essential principles that must be understood to grasp High School Chemistry Test Questions And Answers in its entirety.
 4. In chapter 3, the author will examine the practical applications of High School Chemistry Test Questions And Answers in daily life. This chapter will showcase real-world examples of how High School Chemistry Test Questions And Answers can be effectively utilized in everyday scenarios.
 5. In chapter 4, the author will scrutinize the relevance of High School Chemistry Test Questions And Answers in specific contexts. The fourth chapter will explore how High School Chemistry Test Questions And Answers is applied in specialized fields, such as education, business, and technology.
 6. In chapter 5, the author will draw a conclusion about High School Chemistry Test Questions And Answers. This chapter will summarize the key points that have been discussed throughout the book.
- The book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of High School Chemistry Test Questions And Answers.

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