

17. Assertion : Acetylene is more acidic than ethane.  
Reason : Acetylene has sp character of carbon and therefore more s-character.
18. Assertion : Dipole moment of cis Isomer is less than the trans isomer.  
Reason : cis and trans are the geometrical isomers.

### SECTION-B

- 19.(a) What is the lowest value of n that allows g orbital to exist?  
(b) How many electrons in an atom may have the following quantum numbers?

$$n = 4, m_s = -\frac{1}{2} \quad (2 \times 1 = 2)$$

- 20.(a) What will be the effect on equilibrium of the following reaction, when volume of vessel increases?



What will be the value of Kc for the reverse reaction, if Kc for the decomposition of phosphorus pentachloride is  $8.3 \times 10^{-3}$  (2 \times 1 = 2)

21. Balance the following equation by ion electron method in acidic medium.



22. Draw the resonance structures for the compound  $\text{CH}_2=\text{CH}-\text{C}=\text{O}$



23. What do you mean by buffer solution. Explain the types of buffer with suitable example. (2)
24. Discuss the mechanism of Friedal Craft Acylation of benzene ring. (2)

OR

Draw Newman projections for the conformations of ethane. Which of these conformations is more stable and why? (2)

25. Arrange the elements N, P, O and S in the order of

- (a) Increasing first ionization enthalpy  
(b) Increasing Non-metallic character (2)

# Hbse Chemistry Paper

**Padhraic Smyth**



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### Table of Contents **Hbse Chemistry Paper**

1. Understanding the eBook Hbse Chemistry Paper
  - The Rise of Digital Reading Hbse Chemistry Paper
  - Advantages of eBooks Over Traditional Books
2. Identifying Hbse Chemistry Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Hbse Chemistry Paper
  - User-Friendly Interface
4. Exploring eBook Recommendations from Hbse Chemistry Paper
  - Personalized Recommendations
  - Hbse Chemistry Paper User Reviews and Ratings
  - Hbse Chemistry Paper and Bestseller Lists

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  - Hbse Chemistry Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Hbse Chemistry Paper
  - Highlighting and Note-Taking Hbse Chemistry Paper
  - Interactive Elements Hbse Chemistry Paper
8. Staying Engaged with Hbse Chemistry Paper
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Hbse Chemistry Paper
9. Balancing eBooks and Physical Books Hbse Chemistry Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Hbse Chemistry Paper
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  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
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  - Setting Reading Goals Hbse Chemistry Paper
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Hbse Chemistry Paper
  - Fact-Checking eBook Content of Hbse Chemistry Paper
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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