

Q.41 The vapour pressure of pure liquid B at temperature  $T_0$  is 0.5 atm. The partial pressure of B in the vapour phase that is in equilibrium with the liquid solution consisting of 30 mol% A and 70 mol% B at temperature  $T_0$  is (assume both liquid and vapour phases behave ideally)

- (A) 0.35 atm                      (B) 0.50 atm                      (C) 0.70 atm                      (D) 1.00 atm

Q.42 During low temperature plastic deformation of an under-aged precipitation hardened alloy, dislocations

- (A) climb to completely avoid the precipitate  
(B) loop around the precipitate  
(C) cross-slip to completely avoid the precipitate  
(D) cut through the precipitate

Q.43 According to Hume-Rothery rules, extensive solid solubility between elements X and Y is promoted by the two factors in the following list :

- P. Same crystal structure of X and Y  
Q. Large atomic size difference ( $> 20\%$ ) between X and Y  
R. Same valence of X and Y  
S. Large difference in melting points of X and Y

- (A) P, Q                      (B) P, R                      (C) Q, S                      (D) P, S

Q.44 At constant temperature and pressure, two phases  $\alpha$  and  $\beta$  will be in equilibrium when

- (A) chemical potential of each component is the same in  $\alpha$  and  $\beta$   
(B) partial molar free energy of each component is NOT the same in  $\alpha$  and  $\beta$   
(C) Gibbs free energy of mixing is minimum  
(D) enthalpy of mixing is zero

Q.45 The stress applied on a material is

$$\sigma_y = \begin{bmatrix} 21 & 0 & 0 \\ 0 & 21 & 0 \\ 0 & 0 & 21 \end{bmatrix} \text{ MPa.}$$

The maximum shear stress experienced by it is

- (A) 0 MPa                      (B) 10.5 MPa                      (C) 21 MPa                      (D) 63 MPa

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## **Table of Contents Gate 2009 Question Paper**

1. Understanding the eBook Gate 2009 Question Paper
  - The Rise of Digital Reading Gate 2009 Question Paper
  - Advantages of eBooks Over Traditional Books
2. Identifying Gate 2009 Question Paper
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Gate 2009 Question Paper
  - User-Friendly Interface
4. Exploring eBook Recommendations from Gate 2009 Question Paper
  - Personalized Recommendations
  - Gate 2009 Question Paper User Reviews and Ratings
  - Gate 2009 Question Paper and Bestseller Lists
5. Accessing Gate 2009 Question Paper Free and Paid eBooks
  - Gate 2009 Question Paper Public Domain eBooks
  - Gate 2009 Question Paper eBook Subscription Services
  - Gate 2009 Question Paper Budget-Friendly Options

6. Navigating Gate 2009 Question Paper eBook Formats
  - ePub, PDF, MOBI, and More
  - Gate 2009 Question Paper Compatibility with Devices
  - Gate 2009 Question Paper Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Gate 2009 Question Paper
  - Highlighting and Note-Taking Gate 2009 Question Paper
  - Interactive Elements Gate 2009 Question Paper
8. Staying Engaged with Gate 2009 Question Paper
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Gate 2009 Question Paper
9. Balancing eBooks and Physical Books Gate 2009 Question Paper
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Gate 2009 Question Paper
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Gate 2009 Question Paper
  - Setting Reading Goals Gate 2009 Question Paper
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Gate 2009 Question Paper
  - Fact-Checking eBook Content of Gate 2009 Question Paper
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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