

E1

PACT TON

Volume = 1 unit External pressure = 1 atm Internal pressure = 1 atm Temperature = 200 K

Volume = 2 units

External pressure = 0.50 atm External pressure = 0.33 atm Internal pressure = 0.50 atm Internal pressure = 0.33 atm Temperature = 200 K

Volume = 3 units

E3

Temperature = 200 K

- Consider the gas samples in Model 2.
 - a. Name two materials that the containers in Model 2 could be made from that would ensure that they were "flexible"?

The container could be made of Rubber or latex.

b. What is always true for the external and internal pressures of a gas in a flexible container?

The external and internal pressures are equal in a flexible container

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