Ideal Gas Law & Gas Stoichiometry Quiz SCH3U

Names

Product)
15 5 4 Assessment of Learning

R = 0.31 kPa x L 1 scm = 101.325 kPa = 760 torr = 760 mmHg SATP = 25°C and 100 kPa

Mol x E

STP = 0°C and 101.3 kPa

1. What is the volume, in milliliters, of 0.500 g of G/H, gas at STP? (3T, 1C)

2. A fire extinguisher filled with carbon dioxode has a mass of 2970 g. After releasing all of the CO_b, the mass of the extinguisher is 1235 g. What was the pressure reading on the gauge before any CO_b was released, if the volume of the extinguisher is 3.95 L and it is stored at a temperature of 249C7 (4A, 1C)

 How many liters of CO_b are made if 9.5 L of acetylene (C_bH_b) react in a complete combustion reaction with 1 3.7 L of oxygen at STP?

Gas Stoichiometry Quiz Answers

Ralph Thomas Myers

Gas Stoichiometry Quiz Answers:

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