

WS Gas Laws - KEY

Boyle's Law
 $P_1 \times V_1 = P_2 \times V_2$ At constant temperature

Charles's Law
 $\frac{V_1}{T_1} = \frac{V_2}{T_2}$

At constant pressure
 Temperature in Kelvin
 $^{\circ}\text{C} + 273 = \text{K}$

Lussac's Law
 $\frac{P_1}{T_1} = \frac{P_2}{T_2}$

At constant volume
 Temperature in Kelvin

Combined Gas Law

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2}$$

STP: 273 K & 1.00 atm

Concepts: (circle one choice)

- If the volume of a container of gas is reduced, what will happen to the pressure inside the container?
increase or decrease
- What happens to the volume of a gas when it is cooled to a lower temperature?
increase or decrease
- What happens to the pressure in a rigid container when it is slowly warmed?
increase or decrease

Sample Calculations: (Remember: ALL temperatures **MUST** be in **K**, **NOT** **°C**)

- A sample of gas at 240 K and 670 mmHg occupies a 0.128 L volume. What volume will the gas occupy at 198 K if the pressure remains constant?

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{(0.128 \text{ L})}{(240 \text{ K})} = \frac{V_2}{(198 \text{ K})} \quad \frac{(0.128)(198)}{(240)} = V_2 = 0.106 \text{ L}$$

- A sample of gas is in a steel container at -75.0°C and 1.48 atm. At what temperature will the sample have a pressure of 7.35 atm?

$$K = -75 + 273 = 198 \text{ K}$$

$$\frac{P_1 V_1}{T_1} = \frac{P_2 V_2}{T_2} \quad \frac{(1.48 \text{ atm})}{(198 \text{ K})} = \frac{(7.35 \text{ atm})}{T_2} \quad T_2 = \frac{(7.35)(198)}{(1.48)} = 983 \text{ K}$$

- In an airplane, a gas sample occurs at a volume of 1.50 L at 760 mmHg. Suppose, while flying, the airplane loses pressure and the volume of the gas increases to 11.40 L. What is the pressure in the airplane if the temperature is constant?

$$\frac{P_1 V_1}{X_1} = \frac{P_2 V_2}{X_2} \quad \frac{(760 \text{ mmHg})(1.50 \text{ L})}{(1.50)} = P_2 \frac{(11.40 \text{ L})}{(11.40)} \quad \frac{(760)(1.50)}{(1.50)} = P_2 = 100 \text{ mmHg}$$

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Chemistry: Molecules, Matter, and Change Media Activities Book Loretta Jones, Carl Hoeger, Peter William Atkins, Regina Schoenfeld-Tacher, 2000-01-15 The Media Activity Book MAB for Jones Atkins Chemistry molecules matter and change contains chapters with lists and descriptions of some of the media available as you study the chapter Each activity begin with a specific textbook reference Then you are given a time estimate of how long it will take to use the media An M media icon in the margin of the textbook means that media exists to support that area of text The media is found in three different places on the website and on two CDs

Instructional Technology Research, Design and Development: Lessons from the Field Alias, Nor Aziah, 2011-11-30 Design and development research which has considerable implications for instructional design focuses on designing and exploring products artifacts and models as well as programs activity and curricula Instructional Technology Research Design and Development Lessons from the Field is a practical text on design and development research in the field of instructional technology This book gives readers an overview of design and development research and how it is conducted in different contexts and for various purposes Further this reference source provides readers with practical knowledge on design and development research gained through investigation of lessons learned in the field

Prentice Hall Chemistry, 2000 Activities for a Differentiated Classroom: Level 6 Wendy Conklin, 2011-02-01 Easily implement grade appropriate lessons suitable for Grade 6 classrooms Based on current research these easy to use lessons are based on a variety of strategies to differentiate your instruction Activities are included to allow access to all learners ZIP file contains interactive whiteboard compatible resources including sample projects templates and assessment rubrics This resource is correlated to the Common Core State Standards and is aligned to the interdisciplinary themes from the Partnership for 21st Century Skills

The Three Sisters Ying Wu, John J. Carroll, Yongle Hu, 2019-05-21 This is the seventh volume in the series Advances in Natural Gas Engineering focusing on carbon dioxide CO₂ capture and sequestration acid gas injection and enhanced oil recovery the three sisters of natural gas engineering This volume includes information for both upstream and downstream operations including chapters detailing the most cutting edge techniques in acid gas injection carbon capture chemical and thermodynamic models and much more Written by some of the most well known and respected chemical and process engineers working with natural gas today the chapters in this important volume represent the most state of the art processes and operations being used in the field Not available anywhere else this volume is a must have for any chemical engineer chemist or process engineer in the industry Advances in Natural Gas Engineering is an ongoing series of books meant to form the basis for the working library of any engineer working in natural gas today

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