

# Gas Laws Problem Packet

**Gas Laws Packet**

Name \_\_\_\_\_ Date \_\_\_\_\_ Period \_\_\_\_\_

**Introduction**

Standard Conditions

Unit	Standard Pressure
atm	1.013
mmHg	760
cmHg	76
torr	760

**22.4 L = 1 mol at STP**

• STP (standard temperature and pressure) is 0°C and 1 atm.

• The gas laws:

Gas Law	Equation	Variables
Boyle's Law	$P_1V_1 = P_2V_2$	P, V
Charles's Law	$\frac{V_1}{T_1} = \frac{V_2}{T_2}$	V, T
Gay-Lussac's Law	$\frac{P_1}{T_1} = \frac{P_2}{T_2}$	P, T
Avogadro's Law	$\frac{V_1}{n_1} = \frac{V_2}{n_2}$	V, n
Combined Gas Law	$\frac{P_1V_1}{T_1} = \frac{P_2V_2}{T_2}$	P, V, T
Ideal Gas Law	$PV = nRT$	P, V, n, T

• The constant R used in the ideal gas law depends on the unit of pressure:

- a. Units:  $R = 0.0821 \frac{\text{atm} \cdot \text{L}}{\text{mol} \cdot \text{K}}$
- b. Units:  $R = 62.36 \frac{\text{mmHg} \cdot \text{L}}{\text{mol} \cdot \text{K}}$

**Notes:**

- Temperature must always be in Kelvin:  $K = ^\circ\text{C} + 273$
- If you are given two different units of pressure, convert one.
- Create two columns below (initial and final). Write what you're given in the problem. Mark the two that use the variable you're given, and solve for the unknown. For ideal gas laws, write what you are given, insert those in equation and solve for unknown.
- Avogadro's Law: Solve for volume to use ideal standard values.
- Graham's Law: You will need to know the molecular mass for these problems. Double and triple check you have the right of effusion and molecular masses in the correct units before solving the equation (this is an inverted relationship). You must calculate the square root or square both sides of the equation before solving things.

**Charles' Law  $V_1/T_1 = V_2/T_2$**

1. The temperature of a gas is directly proportional to its volume. If the temperature of a gas is doubled, its volume will also double. If the temperature of a gas is halved, its volume will also halve.

2. A gas at 27°C and 1.0 atm has a volume of 2.0 L. What is the volume of the gas at 127°C and 1.0 atm?

**Graham's Law  $\sqrt{\frac{M_1}{M_2}} = \sqrt{\frac{r_2}{r_1}}$**

1. The rate of effusion of a gas is inversely proportional to the square root of its molecular mass. If the molecular mass of a gas is doubled, the rate of effusion will be reduced by a factor of  $\sqrt{2}$ .

2. A gas with a molecular mass of 16 g/mol effuses at a rate of 1.0 L/min. What is the rate of effusion of a gas with a molecular mass of 64 g/mol?

3. A gas with a molecular mass of 16 g/mol effuses at a rate of 1.0 L/min. What is the rate of effusion of a gas with a molecular mass of 4 g/mol?

4. A gas with a molecular mass of 16 g/mol effuses at a rate of 1.0 L/min. What is the rate of effusion of a gas with a molecular mass of 32 g/mol?

5. A gas with a molecular mass of 16 g/mol effuses at a rate of 1.0 L/min. What is the rate of effusion of a gas with a molecular mass of 8 g/mol?

6. A gas with a molecular mass of 16 g/mol effuses at a rate of 1.0 L/min. What is the rate of effusion of a gas with a molecular mass of 2 g/mol?

7. A gas with a molecular mass of 16 g/mol effuses at a rate of 1.0 L/min. What is the rate of effusion of a gas with a molecular mass of 1 g/mol?

8. A gas with a molecular mass of 16 g/mol effuses at a rate of 1.0 L/min. What is the rate of effusion of a gas with a molecular mass of 0.5 g/mol?

9. A gas with a molecular mass of 16 g/mol effuses at a rate of 1.0 L/min. What is the rate of effusion of a gas with a molecular mass of 0.25 g/mol?

10. A gas with a molecular mass of 16 g/mol effuses at a rate of 1.0 L/min. What is the rate of effusion of a gas with a molecular mass of 0.125 g/mol?

All Gas Laws Included!



# Gas Laws Packet Answers

**Rosalind Malcolm**



## **Gas Laws Packet Answers:**

The Science Teacher's Activity-A-Day, Grades 5-10 Pam Walker, Elaine Wood, 2010-10-05 A hands on and fun filled resource for teaching science to middle and high school students New in the 5 Minute Fundamentals Series The Science Teacher's Activity A Day Grades 6 12 includes 180 easy five minute hook or sponge activities to capture learners attention and introduce lessons Divided into three units Physical Science Life Science and Earth and Space Science the activities cover topics based on the National Science Education Standards All the book's activities can be done with materials that are inexpensive and easy to find Includes quick and fun sponge activities that are designed to engage students All the activities take about 5 minutes to complete The Science Teacher's Activity a Day is an ideal resource for middle and high school science teachers

**The Handy Chemistry Answer Book** Justin P. Lomont, Ian C. Stewart, 2013-10-01 Don't be mixed up about chemistry Simplify the complex chemical reactions that take place everywhere in our lives with this engaging easy to follow question and answer guide Where would we be without atoms and compounds Gas liquids solids and plasma Acids and bases Bonds and reactions Matter and energy The Handy Chemistry Answer Book covers the building blocks of life and the universe The secret life of atoms how polar bears aren't actually white why oil and water don't mix and much much more are revealed and explained This informative guide covers the basics of chemistry history atomic structures chemical bonds and reactions organic and inorganic chemistry to more advanced material nuclear chemistry biochemistry physical and theoretical chemistry by answering nearly 1 000 common chemistry questions including What causes lightning How does photosynthesis work What are hard and soft Lewis acids and bases What makes a fabric waterproof What are the twelve principles of green chemistry When did alchemists finally abandon trying to make gold What is Le Chatelier's principle What do the different octane ratings mean at the gas pump What is genetic engineering Why is calcium important for strong bones What is the 18 electron rule Why does chocolate turn white as it ages Chemical reactions that rule the world their properties structure composition behavior and history are tackled and explained in plain English in The Handy Chemistry Answer Book With many photos illustrations a few formulas molecular diagrams and other graphics this fun fact filled tome is richly illustrated A history of chemistry timeline appendices on Nobel Prize in Chemistry winners a bibliography further reading section glossary of terms a table of physical constants a table of conversion factors and extensive index add to its usefulness

Concentrate Questions and Answers Land Law Rosalind Malcolm, 2018-07-26 The Concentrate Q the content is exceptional Wendy Chinenye Akaigwe London Metropolitan University I would not hesitate to recommend this book to a friend Blessing Denhere Coventry University

**Instructional Technology Research, Design and Development: Lessons from the Field** Alias, Nor Aziah, 2011-11-30 Design and development research which has considerable implications for instructional design focuses on designing and exploring products artifacts and models as well as programs activity and curricula Instructional Technology Research Design and Development Lessons from the Field is a practical text on design and

development research in the field of instructional technology This book gives readers an overview of design and development research and how it is conducted in different contexts and for various purposes Further this reference source provides readers with practical knowledge on design and development research gained through investigation of lessons learned in the field

**Proposed Changes to Natural Gas Laws** United States. Congress. House. Committee on Energy and Commerce. Subcommittee on Fossil and Synthetic Fuels,1983

*Learning and Leading with Technology* ,1996

Applied

Thermodynamics for Meteorologists Sam Miller,2015-06-04 This textbook on atmospheric thermodynamics is for students of meteorology or atmospheric science It also serves as a reference text for working professionals in meteorology and weather forecasting It is unique because it provides complete calculus based derivations of basic physics from first principles and connects mathematical relationships to real world practical weather forecasting applications Worked examples and practice problems are included throughout

*Atmospheric Thermodynamics* Gerald R. North,Tatiana L. Erukhimova,2009-04-09 This textbook presents a uniquely integrated approach in linking both physics and chemistry to the study of atmospheric thermodynamics The book explains the classical laws of thermodynamics focuses on various fluid systems and recognising the increasing importance of chemistry in the meteorological and climate sciences devotes a chapter to chemical thermodynamics which includes an overview of photochemistry Although students are expected to have some background knowledge of calculus general chemistry and classical physics the book provides set aside refresher boxes as useful reminders It contains over 100 diagrams and graphs to supplement the discussions and a similar number of worked examples and exercises with solutions included at the end of the book It is ideal for a single semester advanced course on atmospheric thermodynamics and will prepare students for higher level synoptic and dynamics courses

Cases and Materials on Oil and Gas Law John S. Lowe,2002 The pages of this book are filled with theoretical enigmas that intrigue legal theorists yet oil and gas law is practical enough to satisfy the desire for real law shared by most second and third year law students This edition contains numerous changes from the first two editions The authors have added materials on environmental law in a new Chapter 9 They have also inserted the materials on implied covenants into Chapter 2 alongside express lease provisions and moved the materials on royalty disputes into Chapter 3 They have updated and revised all of the chapters

*Meteorology* ,1971

*United States Government Organization Manual* ,1969

*Terrestrial Hydrometeorology* W.

James Shuttleworth,2012-01-30 Both hydrologists and meteorologists need to speak a common scientific language and this has given rise to the new scientific discipline of hydrometeorology which deals with the transfer of water and energy across the land atmosphere interface Terrestrial Hydrometeorology is the first graduate level text with sufficient breadth and depth to be used in hydrology departments to teach relevant aspects of meteorology and in meteorological departments to teach relevant aspects of hydrology and to serve as an introductory text to teach the emerging discipline of hydrometeorology The book will be essential reading for graduate students studying surface water hydrology meteorology and hydrometeorology It

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hydrometeorology      **Resources in Education** ,1993      **Cases and Materials on Oil and Gas Law** ,1998      *A First Course in Atmospheric Thermodynamics* Grant W. Petty,2008-01-01 This textbook is written for meteorology majors who require an initial introduction to the physical properties of the atmosphere and to the essential principles and real world applications of atmospheric thermodynamics These topics are supplemented by a sampling of techniques and technologies related to atmospheric measurements and observations A unique tutorial included as an appendix teaches students how to attack physical problems symbolically deferring numerical calculations until the final step in the solution The author s objectives include not only covering the traditional core subject matter of an undergraduate thermodynamics course but also facilitating students transition from a purely abstract understanding of calculus and physics concepts to the confident application of both to the science of meteorology      **Examining Impacts of Federal Natural Resources Laws Gone Astray** United States. Congress. House. Committee on Natural Resources. Subcommittee on Oversight and Investigations,2017      **Examining Impacts of Federal Natural Resources Laws Gone Astray, Part II, Serial No. 115-16, July 18, 2017, 115-1** ,2017      **Business Law** Dr. S. K. Singh, Dr. Angad Tiwary, Nikhil Gupta,2021-05-31 Unit I Indian Contract Act 1872 1 Business Mercantile Law An Introduction 2 Indian Contract Act 1872 An Introduction 3 Contract Meaning Definition and Characteristics of a Valid Contract 4 Agreement Meaning Kinds and Difference 5 Proposal Offer Acceptance Communication and Revocation 6 Capacity of Parties to Contract or Parties Competency to Contract 7 Free Consent 8 Lawful Consideration and Objects 9 Agreements Expressly Declared as Void 10 Contingent Contracts 11 Performance of Contracts and Appropriation of Payments 12 Discharge of Contracts 13 Quasi or Implied Contracts of Certain Relations Resembling those Created by Contracts Sections 68 to 72 14 Remedies for Breach of Contract Unit II Special Contracts 15 Contract of Indemnity and Guarantee 16 Contract of Bailment and Pledge 17 Contracts of Agency Unit III The Sale of Goods Act 1930 18 The Sale of Goods Act 1930 An Introduction 19 Conditions and Warranties 20 Effects of the Contract of Sale Transfer of Ownership and Title 21 Performance of Contract of Sale 22 Remedial Measures and Auction Sale Unit IV The Negotiable Instruments Act 1881 1 Negotiable Instruments Act 1881 Introduction 2 Parties to a Negotiable Instruments 3 Negotiation 4 Presentment and Dishonour of Negotiable Instruments 5 Discharge of Negotiable Instruments 6 Hundis 7 Banker and Customer Unit V G S T G S T Format and Computing Process      The Law Reports ,1909      **Activities of Federal Law Enforcement Agencies Toward the Branch Davidians** United States. Congress. House. Committee on the Judiciary. Subcommittee on Crime,1996

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