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Chris Redston, Gillie Cunningham

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web in the solubility and temperature gizmo you will study how temperature affects how much solute will dissolve in a solution to begin check that potassium nitrate is selected and the temp of the water is 20 c

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web jun 4 2019 the solubility of the solution is equal to the maximum concentration of the solute what is the solubility of potassium nitrate in 20 c water 3 experiment click reset and select sodium chloride with the temp still set to 20 c click ok add sodium chloride to the beaker until it starts piling up at the bottom

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web solution if you can't see the sugar how can you tell that it is there does sugar dissolve more easily in hot water or cold water gizmo warm up a solution generally consists of two parts a solute that is dissolved and a solvent that the solute is dissolved into for example sugar is a solute that is dissolved into the solvent water in the

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web 1 in this solution what is the solute potassium nitrate what is the solvent water 2 click add 10 g to mix 10 g of potassium nitrate into the water a did all of the potassium nitrate dissolve b how can you tell yes all of the potassium nitrate disappeared as it was stirred into the water activity a solubility

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web feb 15 2023 correct answer a substance a is less soluble in water than substance b which one of the following graphs best describes the solubility of potassium nitrate KNO_3 in water at different temperatures

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