Solubility and Temperature

Answer Key

Vocabulary: concentration, dissolve, homogeneous mixture, solubility, solubility curve, solute, solution, solvent

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

[Note: The purpose of these questions is to activate prior knowledge and get students thinking. Students are not expected to know the answers to the Prior Knowledge Questions.]

What happens when you stir a spoonful of sugar into hot water?

The sugar dissolves into the water.

When sugar or another substance is dissolved in water, it disappears from view and forms a homogeneous mixture with the water, also called a solution.

If you can't see the sugar, how can you tell that it is there?

The water tastes sweet and may appear slightly cloudy. [You can boil or evaporate the water to see solid sugar again.]

Does sugar dissolve more easily in hot water or cold water? Hot water

Gizmo Warm-up

A solution generally consists of two parts, a **solute** that is dissolved and a **solvent** that the solute is dissolved into. For example, sugar is a solute that is dissolved into the solvent water. In the Solubility and Temperature GizmoTM, you will study how temperature affects how much solute will dissolve in a solution.

To begin, check that Potassium nitrate is selected and the Temp, of the water is 20 °C. Click OK.

- 1. In this solution, what is the solute? Potassium nitrate
- And 1 g And 10 g And 20 g
- What is the solvent? Water
- Click Add 10 g to mix 10 g of potassium nitrate into the water.
 - A. Did all of the potassium nitrate dissolve? Yes
 - B. How can you tell?

All of the potassium nitrate disappeared as it was stirred into the water.

Explore Learning Answer Key Solubility And Temperature

Scott C. Dulebohn

Explore Learning Answer Key Solubility And Temperature:

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