

and scattering is *incorrect*:

- Negatively charged object .
- Increase in the impact parameter .
- Coulomb interaction with the outer valence 4-d

es are deflected through  $45^\circ$  or more by a  $0.1 \mu\text{m}$  gold

**A MILLIA ISLAMIA**  
[11-12-13]

obtained in the  $H_\alpha$  emission spectrum from hydrogen

$\lambda$ ) of an atomic spectral line of wavelength 500 nm

correctly describes a system of two identical, non-  
and 'b' ?

$$\begin{aligned}\psi_A &= 1/2[\psi_a(1)\psi_b(2) - \psi_a(2)\psi_b(1)] \\ &= [\psi_a(1)\psi_b(2) - \psi_a(2)\psi_b(1)] \\ \psi_A &= 1/\sqrt{2}[\psi_a(1)\psi_b(2) - \psi_a(2)\psi_b(1)] \\ \psi_A &= 1/\sqrt{2}[\psi_a(1)\psi_b(2) + \psi_a(2)\psi_b(1)]\end{aligned}$$

described by the term symbol  $3^2S_{1/2}$ . What are the  
e valence electron?

65. Which of the following statements

- (A) Molecular rotational states are
- (B) The selection rule for molecu-
- (C) Hydrogen molecule ( $H_2$ ) does
- (D) Transitions between low lying

**887 1A**

66. A 50 kW broadcasting antenna em  
photons are emitted per second are

- (A)  $1.5 \times 10^{31}$
- (B)  $1.7 \times 10^{31}$
- (C)  $1.9 \times 10^{31}$
- (D)  $2 \times 10^{31}$

67. A photon is incident on an electros  
observed to be 5 times the wavelength  
the wavelength of the incident photon

- (A)  $2 \times 10^{-13}\text{m}$
- (B)  $3 \times 10^{-13}\text{m}$
- (C)  $4 \times 10^{-13}\text{m}$
- (D)  $5 \times 10^{-13}\text{m}$

68. The de Broglie wavelength of a pa  
box of length L, is

- (A) 4L
- (B) 2L
- (C) L
- (D)  $L/2$

69. The operator  $\left(\frac{d}{dx} + x\right)\left(\frac{d}{dx} - x\right)$  is

- (A)  $\frac{d^2}{dx^2} - x^2$
- (B)  $\frac{d^2}{dx^2} - x^2 - 1$
- (C)  $\frac{d^2}{dx^2} - 2x\frac{d}{dx} + 1$
- (D)  $\frac{d^2}{dx^2} - 2x\frac{d}{dx} - x^2$

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**Wolfgang Guggemos**



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