(b) Ethanoic acid reacts with another alcohol, Y, to produce an ester Z.

(i) Alcohol Y has molar mass 74 g mol<sup>-1</sup> and the following composition by mass:

carbon, C = 64.9%

hydrogen, H = 13.5%

oxygen, O = 21.6%.

Use all these data to confirm that the molecular formula for  $\mathbf{Y}$  is  $C_oH_{1o}O$ . Show your working.

(2)

(iii) Draw the displayed formulae of the four possible structures of alcohol Y.

(2)

Alcohol 1	Alcohol 2
Alcohol 3	Alcohol 4

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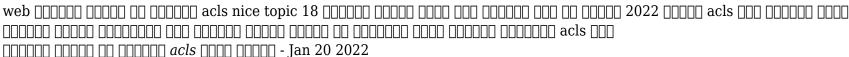
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